

THE EMERGING ISSUE

PER- & POLYFLUOROALKYL SUBSTANCES (PFASs)

Big Picture, Challenges and Solutions

Ian Ross, Ph.D.

Global PFASs Lead
Arcadis

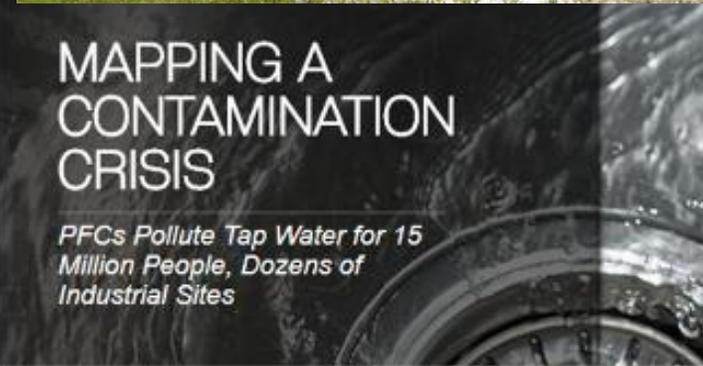
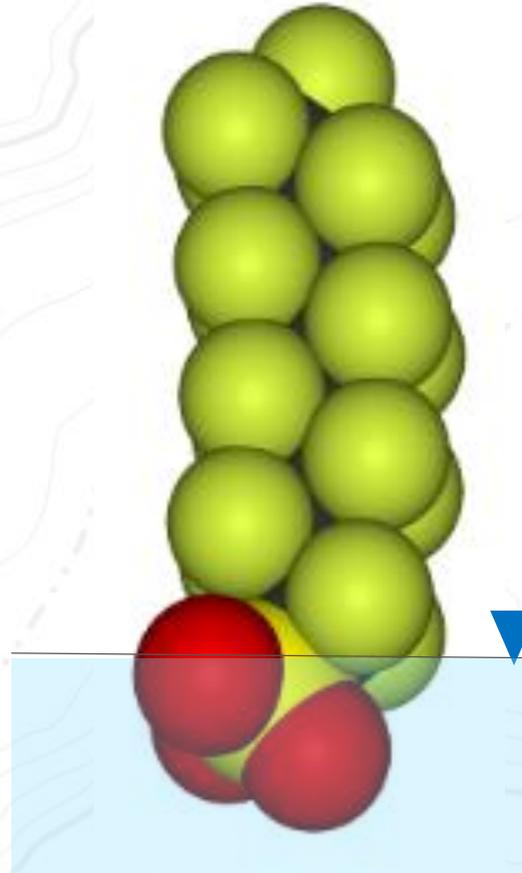
PFASs

Contents

- PFASs News
- PFASs Chemistry
- Replacement Chemistry
- Regulatory Evolution
- State of UK waters
- Ingenious Treatment Solutions
- Conclusions

PFASs

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PFASs Introduction



PFAS comprises many thousands of compounds –multiple sources



Advanced analytical methods are being adopted to measure PFAS



PFAS are impacting drinking water worldwide



None of the PFASs biodegrade, some biotransform to daughter compounds that are extremely persistent



Some PFAS are classed as persistent organic pollutants



Dramatically increasing regulatory concern

Unique Characteristics of PFASs

Longer chain PFASs stick their tails in the air -the longer perfluoroalkyl chains migrate to the gas:liquid interface, so stratify in solution

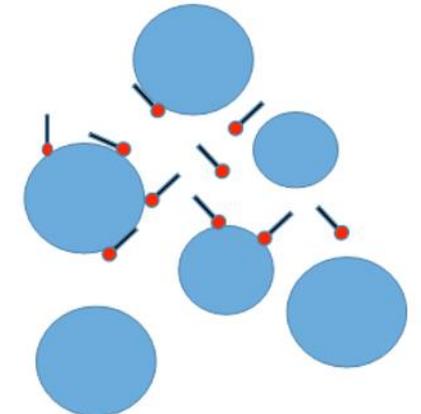
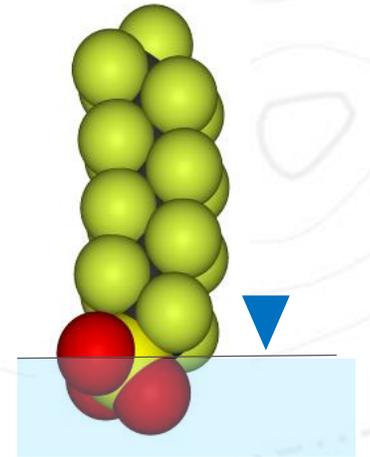
PFASs bioaccumulate via interaction with proteins (not fats) -PFASs bind to β -lipoproteins, as mistaken for fatty acids

PFASs tend to be soluble in water -PFASs can be very mobile in the environment as water soluble, unlike most other POPs.

Long chain PFASs are initially excreted in humans then reabsorbed -humans have one the highest levels of renal reabsorption so fail to excrete long chain PFASs, whereas monkeys, mice and rats can excrete at faster rates

PFASs tend to stick together -long chain PFAAs have been identified as layers on surfaces agglomerating by via “molecular brush”

Increased concentrations of PFAAs observed at WWTP outflow vs inlet – PFAA precursors are transformed in municipal / biological / oxidative treatment processes





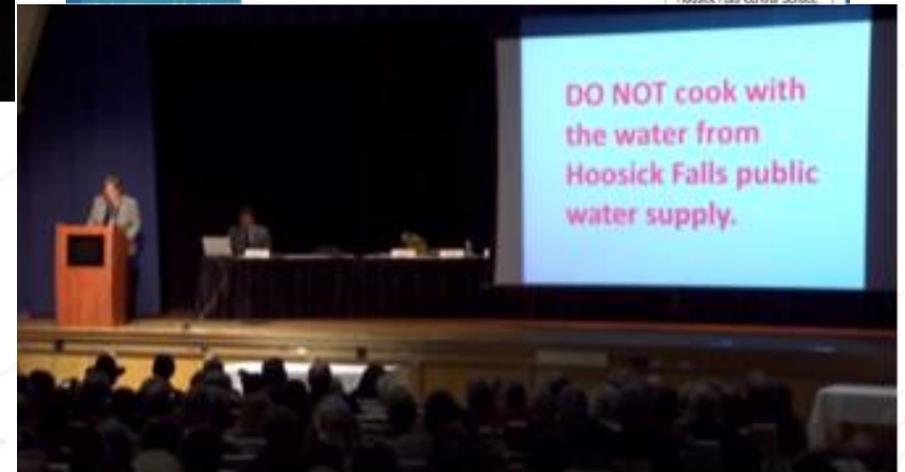
The FDA Just Banned These Chemicals in Food. Are They the Tip of the Iceberg?

FDA banned three toxic food packaging, banning seven cancer-causing food flavor additives say the process highlights flaws



These Chemicals in Pizza Boxes and Carpeting Last Forever

More than 200 scientists around the world document the threats of perfluorinated compounds and call for more government control.



Toxic chemicals in outdoor products of leading brands, Greenpeace study finds

Environment group calls on outdoor clothing companies to phase out PFCA, which have been linked to reproductive and developmental problems



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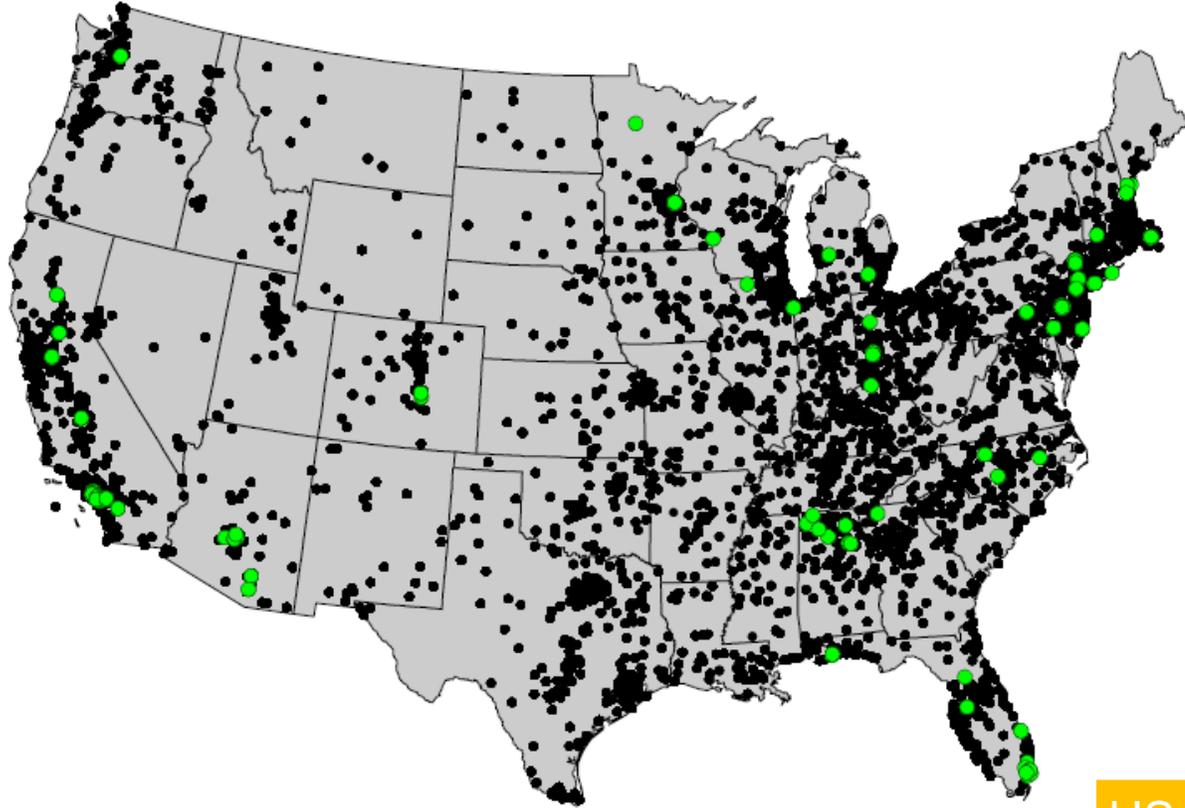
Energy and Environment

Researchers find unsafe levels of industrial chemicals in drinking water of 6 million Americans

By Brady Dennis August 9

Detections of PFAS in drinking water has caused spiraling regulatory concern

PFASs in US Public Water Supplies



USEPA UMCR 3, May 2016

PFASs

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Detected in ~ 2% of large public water supplies

News > World > Americas

Six million Americans drinking water containing unsafe levels of unregulated chemicals, study finds

In one Delaware town, the levels of one such chemical in the water supply were 25 times higher than the EPA deems safe

Tim Walker US Correspondent | @timwalker | Tuesday 9 August 2016 22:57 BST |



US EPA has established the drinking water health advisory levels at 70 ng/L for PFOA/PFOS 19th May 2016

<https://www.epa.gov/ground-water-and-drinking-water/drinking-water-health-advisories-pfoa-and-pfos>

US ATSDR releases 'suppressed' PFAS tox profile

Study confirms EPA guidelines 'woefully underestimate risk', says NGO

21 June 2018 / PFAs, Toxicology, United States

The US Agency for Toxic Substances and Disease Registry has released a controversial draft toxicological profile on four per- and polyfluoroalkyl substances (PFASs). The move comes amid uproar over allegations that other federal agencies were suppressing its release.

Last month, internal EPA emails released under a public records request showed concern that the ATSDR was planning to publish a study with minimal risk levels (MRLs) for the PFASs far below those set by the EPA. One White House staffer feared this would result in a "public relations nightmare".



The Lawyer Who Became Worst Nightmare

Corporate defense attorney for eight years. Then he took on an environmental suit that would end his entire career — and expose a brazen, decades-long history of chemical pollution.

http://www.nytimes.com/2016/01/10/magazine/the-lawyer-who-became-duponts-worst-nightmare.html?_r=0



Erin Brockovich: It's Not Just Flint—America Has a Scary Water Problem

Erin Brockovich and Ken Cook Feb. 8, 2018

Erin Brockovich is a consumer advocate, and Ken Cook is president of the Environmental Working Group.

We must reform our broken chemical laws to prevent more tragedies

Most Americans take our drinking water for granted: turn the tap, fill a glass and drink. Only when a community's health and safety are imperiled do we pay heed to the threat of industrial chemicals in our water supplies. The answer



Toxic Secrets: Where the sites with PFAS contamination are near you

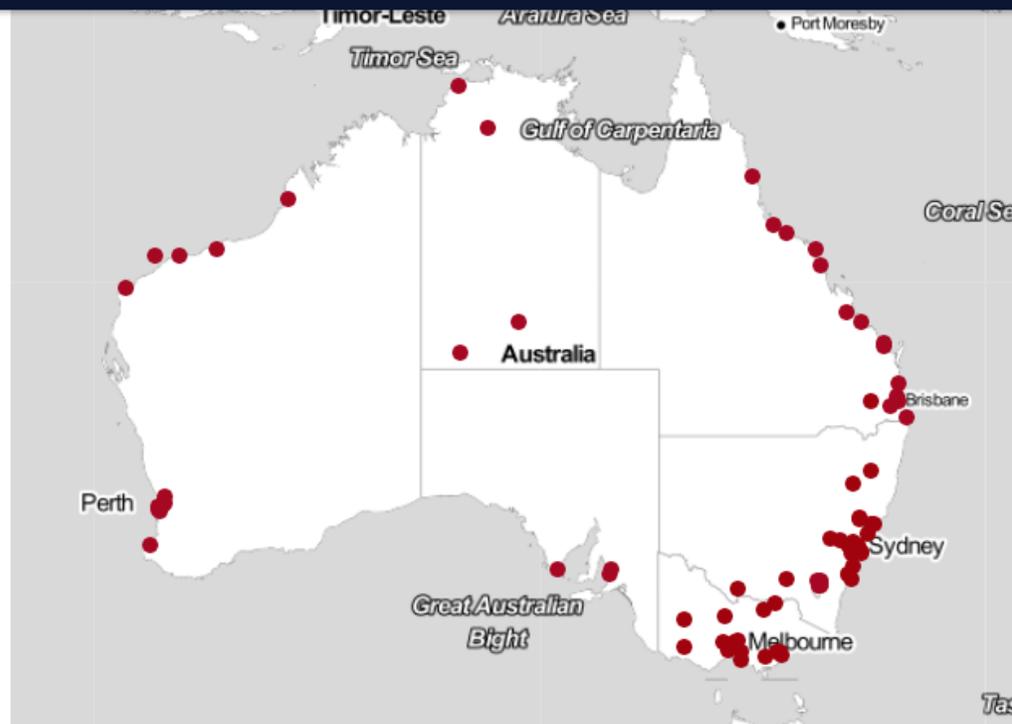
Sites under investigation

By Carrie Fellner & Patrick Begley

17 JUNE 2018

At least 90 sites across Australia are under investigation for elevated levels of per- and poly-fluoroalkyl [PFAS] chemicals.

The Sydney Morning Herald



PFASs

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<https://www.smh.com.au/world/north-america/toxic-secrets-the-town-that-3m-built-where-kids-are-dying-of-cancer-20180613-p4z183.html>

<https://www.smh.com.au/national/report-into-toxic-chemicals-finds-pfas-worse-than-thought-20180621-p4zmy8.html>

<https://www.smh.com.au/lifestyle/health-and-wellness/toxic-secrets-professor-bragged-about-burying-bad-science-on-3m-chemicals-20180615-p4zisc.html>

<https://www.smh.com.au/national/nsw/toxic-secrets-where-the-sites-with-pfas-contamination-are-near-you-20180616-p4z1xc.html>



Toxic firefighting chemicals 'the most seminal public health challenge'

US environmental official says Pfas chemicals found in firefighting foam is contaminating water supplies

Christopher Knaus

@knausc

Wed 18 Oct 2017 00:52 EDT



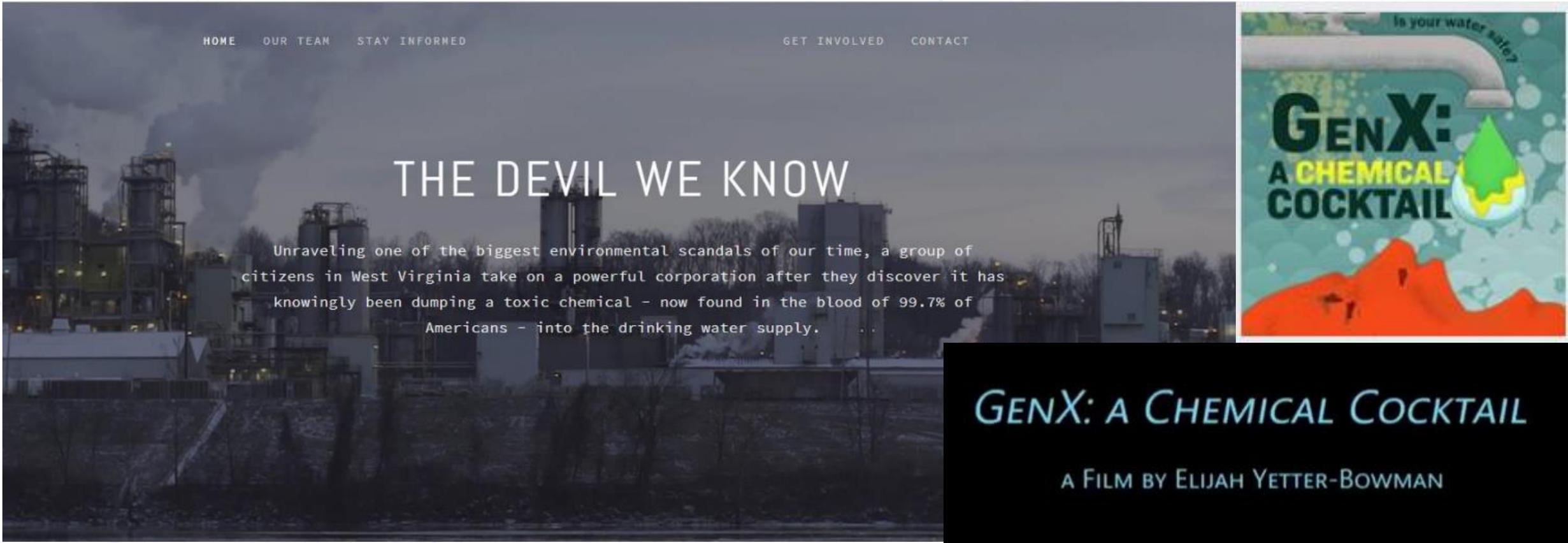
Centers for Disease Control and Prevention
CDC 24/7: Saving Lives, Protecting People™

Patrick Breysse, Director of the CDC's National Centre for Environmental Health, described the chemicals as “one of the most seminal public health challenge for the next decades”

Breysse estimated 10 million Americans were currently drinking contaminated water.

He said soon “we think that hundreds of millions of Americans will be drinking water with levels of these chemicals above levels of concern”

Coming Soon...



The image shows a screenshot of a website and a poster. The website screenshot features a dark background with a factory scene. At the top, there are navigation links: HOME, OUR TEAM, STAY INFORMED, GET INVOLVED, and CONTACT. The main heading is 'THE DEVIL WE KNOW'. Below it, a paragraph reads: 'Unraveling one of the biggest environmental scandals of our time, a group of citizens in West Virginia take on a powerful corporation after they discover it has knowingly been dumping a toxic chemical - now found in the blood of 99.7% of Americans - into the drinking water supply.' To the right is a poster for the film 'GENX: A CHEMICAL COCKTAIL'. The poster has a green and red color scheme and features a water tap with a green and yellow drop. The text on the poster includes 'Is your water safe?' and 'GENX: A CHEMICAL COCKTAIL'. Below the poster is a black box with the text 'GENX: A CHEMICAL COCKTAIL' and 'A FILM BY ELIJAH YETTER-BOWMAN'.

<https://www.thedevilweknow.com/>

<https://www.genxthefilm.org/>

PFASs

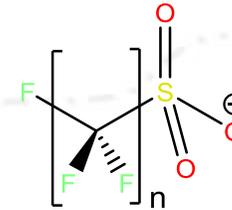
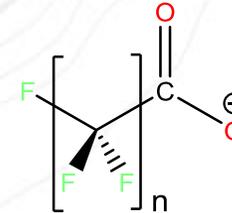
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Perfluoroalkyl Acids (PFAAs)

- **Perfluoroalkyl acids (PFAAs)** previously termed Perfluorinated Compounds (PFCs) generally are the and include:
 - Perfluoroalkyl carboxylates (PFCAs) e.g. PFOA
 - Perfluoroalkyl sulfonates (PFSAs) e.g. PFOS
 - Perfluoroalkyl phosphinic acids (PFPiS); perfluoroalkyl phosphonic acids (PFPAs)
 - Perfluoroalkyl ethers e.g. GenX
- There are many PFAAs with differing chain lengths (generally C1-C18)

Perfluoroalkyl acid naming references the number of fluorinated (n) carbons

- C1 Methane
- C2 Ethane
- C3 Propane
- C4 Butane
- C5 Pentane
- C6 Hexane
- C7 Heptane
- C8 Octane
- C9 Nonane
- C10 Decane
- C11 Undecane
- C12 Dodecane



Perfluoroalkyl Carboxylates (PFCA)*

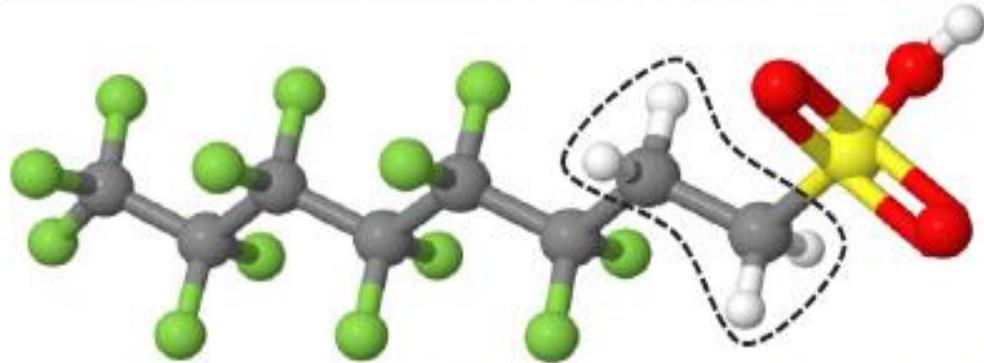
- n=1 Perfluoroethanoic acid (PF_{Et}A)
- n=2 Perfluoropropanoic acid (PF_{Pr}A)
- n=3 Perfluorobutanoic acid (PF_{Ba}A)
- n=4 Perfluoropentanoic acid (PF_{Pe}A)
- n=5 Perfluorohexanoic acid (PF_{Hx}A)
- n=6 Perfluoroheptanoic acid (PF_{Hp}A)
- n=7 Perfluorooctanoic acid (PFOA)
- n=8 Perfluorononanoic acid (PFNA)
- n=10 Perfluorodecanoic acid (PFDA)
- n=11 Perfluoroundecanoic acid (PF_{Ud}A)
- n=12 Perfluorododecanoic acid (PF_{Do}A)

Perfluoroalkyl Sulfonates (PFSA)

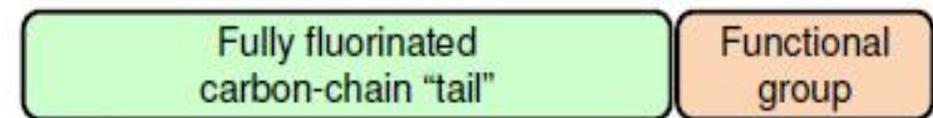
- n=1 Perfluoromethanesulfonic acid (PF_{Me}S)
- n=2 Perfluoroethanesulfonic acid (PF_{Et}S)
- n=3 Perfluoropropanesulfonic acid (PF_{Pr}S)
- n=4 Perfluorobutanesulfonic acid (PF_{Bs}S)
- n=5 Perfluoropentanesulfonic acid (PF_{Pe}S)
- n=6 Perfluorohexanesulfonic acid (PF_{Hx}S)
- n=7 Perfluoroheptanesulfonic acid (PF_{Hp}S)
- n=8 Perfluorooctanesulfonic acid (PFOS)
- n=9 Perfluoronanesulfonic acid (PFNS)
- n=10 Perfluorodecanesulfonic acid (PFDS)
- n=11 Perfluoroundecanesulfonic acid (PF_{Ud}S)
- n=12 Perfluorododecanesulfonic acid (PF_{Do}S)

* The final carbon atom in PFCA's unlike in PFSA's is part of the head group and not fluorinated.

Polyfluorinated Compounds – PFAA Precursors



e.g. Poly-fluorinated - 6:2 Fluorotelomer sulfonate (6:2FTS)



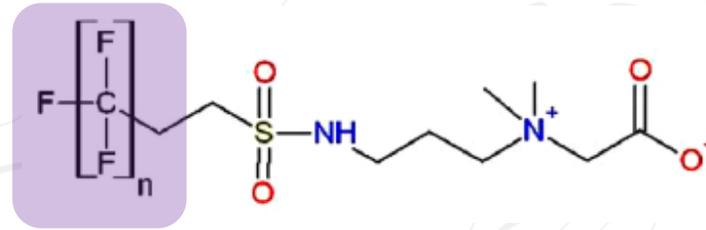
e.g. Per-fluorinated - PFOS

Atoms—Green=Fluorine, Grey=Carbon, White=Hydrogen, Red=Oxygen, Yellow=Sulphur, non-fluorinated carbon atoms circled

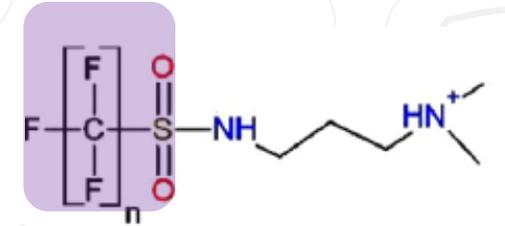
Perfluoroalkyl group –the forever functional group



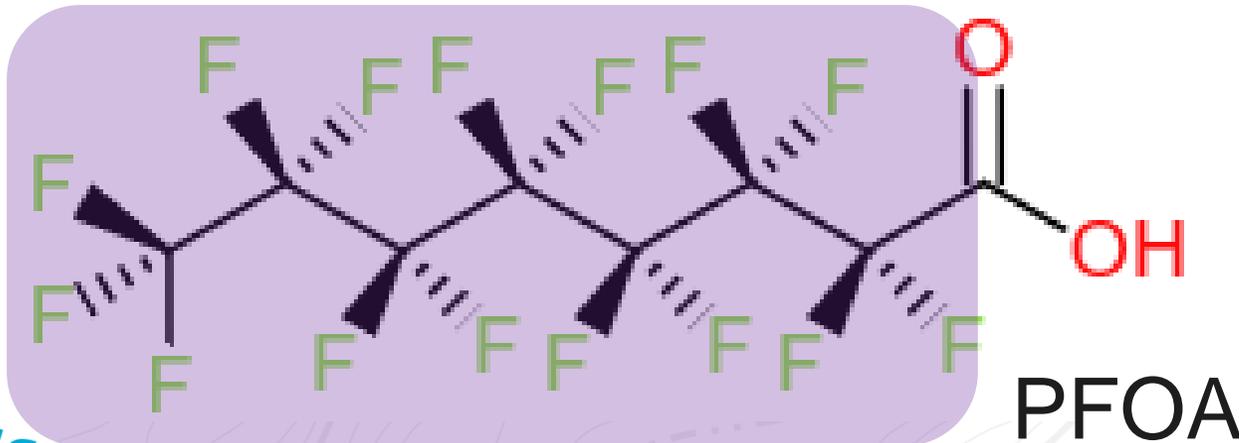
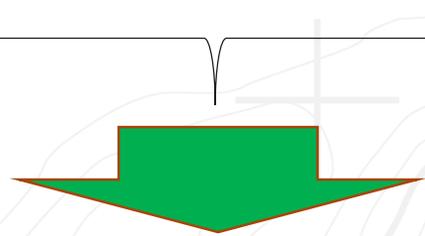
Fluorotelomer alcohol, 8:2 FTOH



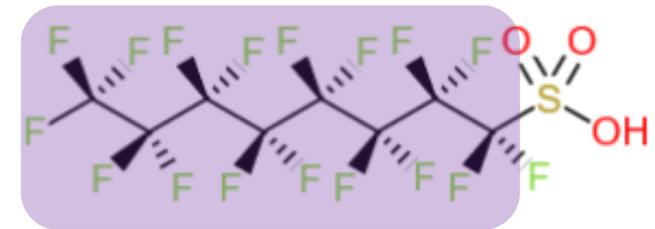
Fluorotelomer Sulfonamido Betaines



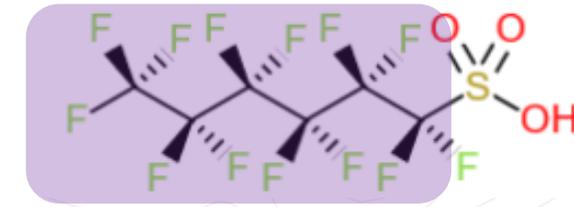
Perfluoroalkyl Sulfonamido Amines



PFOA



PFOS



PFHxS

Poly- and Perfluoroalkyl Substances (PFASs)

More Commonly Regulated

Polyfluorinated
compounds (~5,000
compounds)

Perfluorinated Compounds (PFCs) aka
Perfluoroalkyl Acids (PFAAs)
~25 common individual compounds
but ~100's compounds
PFOS, PFOA, PFHxS, PFBA, GenX

Microbial / Higher Organism Biotransformation

Discovery of 40 Classes of Per- and Polyfluoroalkyl Substances in Historical Aqueous Film-Forming Foams (AFFFs) and AFFF-Impacted Groundwater

Krista A. Barzen-Hanson,[†] Simon C. Roberts,^{∇,‡} Sarah Choyke,[§] Karl Oetjen,[‡] Alan McAlees,^{||} Nicole Riddell,^{||} Robert McCrindle,[⊥] P. Lee Ferguson,[§] Christopher P. Higgins,^{*,‡} and Jennifer A. Field^{*,#}

Class Number	Structure	$n^{a,b}$	Acronym ^c	Confidence Level ^{d,e}	AFFF/CP Found In
21		3-9	n-FSS-PFAS	2b	A, B, C, D, E, G, M, N
22		6-8	n+-FSS-PFAS	3	M, N
23		1-10	UPFAS ^h	3	A, B, C, D, E, M, N, P
24		1-6	H-UPFAS ^h	3	A, B, C, D, E, F, G, M
25		0-8	H-PFAS ^h	3	A, B, C, D, E, F, G, M, N, P
26		5, 7	α -1 PFAS ^h	3	A, B, C, D, E, F, G, M, N, P

13		3-8	N-TAmP-FASA	3	A, B, C, D, E, F, G
14		3-6	N-TAmP-FASAP	3	D, E, F, G
15		4-6	N-CMAmP-FASAP	2b	D, E, F, G
16		3-6	N-CMAmP-FASA	2b	D, E, F, G
17		6, 8, 10	CMAmE-FA	2b	L
18		4, 6, 8	CMAmE-FA	3	L
19	$C_{10}H_{19}O_2SN_2F_{2n+1}$	6, 8, 10	Not applicable	4	1, 2
20	$C_{10}H_{19}O_2SN_2F_{2n+1}$ or $C_{10}H_{17}O_2SN_2F_{2n+1}$	Unknown	Not applicable	5	1, 2

Class Number	Structure	$n^{a,b}$	Acronym ^c	Confidence Level ^{d,e}	AFFF/CP Found In
1		3-6	N-SP-FASA	2b	B, C
2		3-8	N-SPAmP-FASA	2b	A, B, C, F
3		3-9	N-SHOAmP-FASA	3	C, D, E, F, G
4		4-6	N-SHOEAmP-FASA	3	B, C
5		3-8	N-SPAmP-FASAPS	2b	A, B, C
6		3-6	N-dHOPAmEOb-FASA	3	B, C, O
7		2-6	N-dHOPAmEOb-FASAPS	3	A, B, C
8		2-8	N-HOEAmP-FASAPS	2b	A, B, C
9		2-8	N-HOEAmP-FASE	2b	A, B, C, D, E
10		4-6	N-HOEAmP-FASA	3	B, C
11		2-8	N-HOEAmP-FASA	2b	A, B, C, D, E
12		4-8	N-TAmP-N-MeFASA	3	B

Digest AFFF precursors and measure the hidden mass: TOP Assay

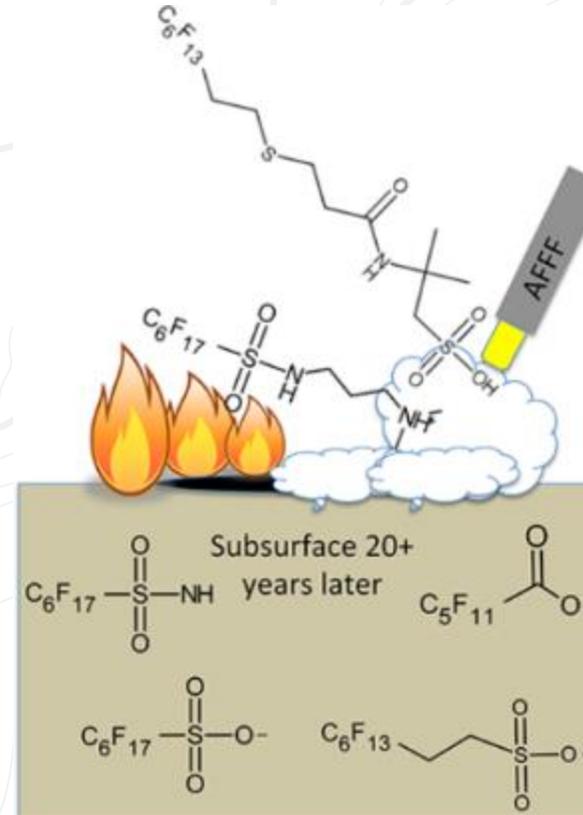
Microbes slowly make simpler PFAA's (e.g. PFOS / PFOA) from PFAS (PFAA precursors) over 20+ years

Need to determine precursor concentrations as they will form PFAAs

Too many PFAS compounds and precursors –so very expensive analysis

Oxidative digest convert PFAA precursors to PFAA's

Indirectly measure precursors as a result of the increased PFAAs formed

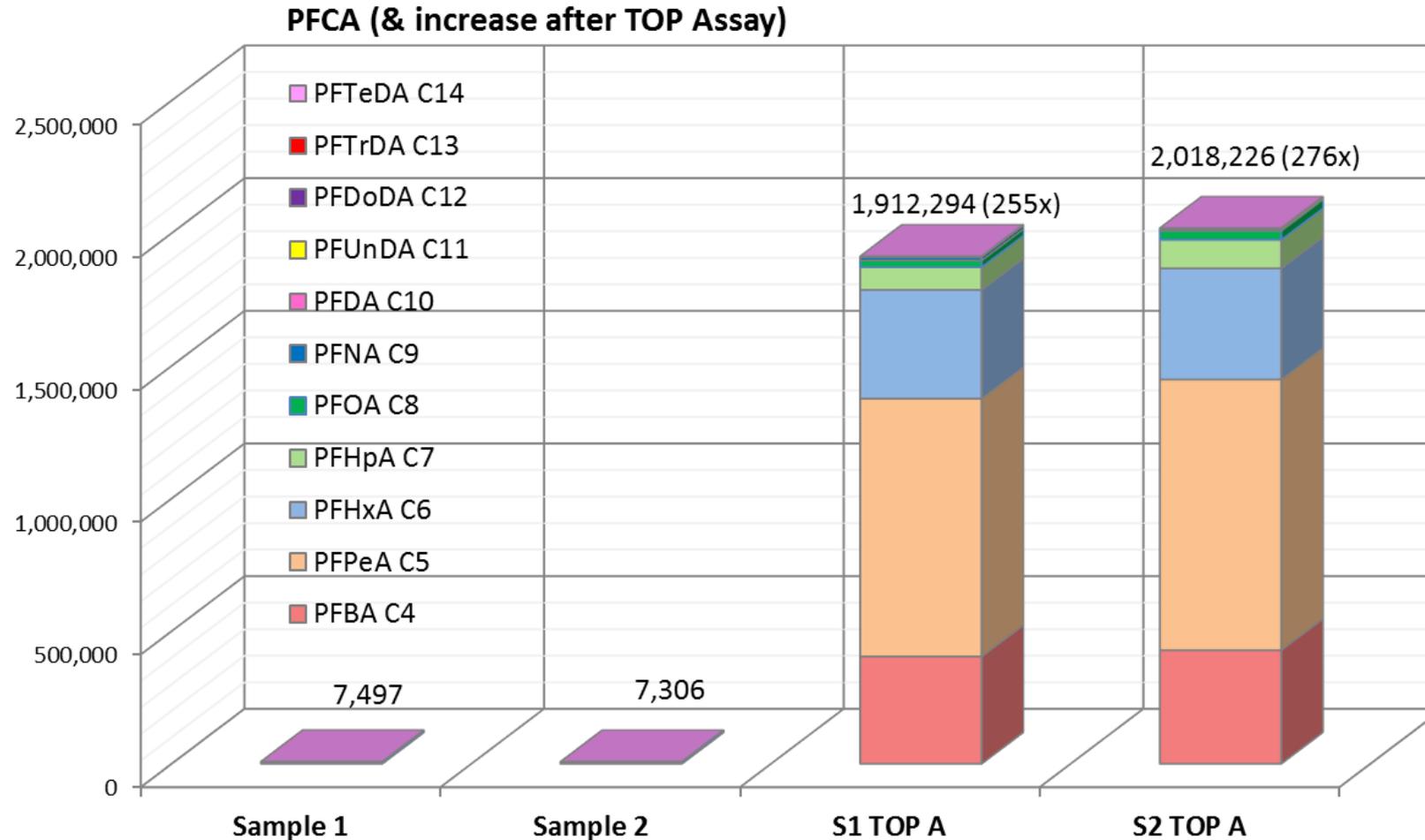


Persistence of Perfluoroalkyl Acid Precursors in AFFF-Impacted Groundwater and Soil

Erika F. Houtz,[†] Christopher P. Higgins,[‡] Jennifer A. Field,[§] and David L. Sedlak^{†,*}



TOP Assay Applied to Surface Water from Recent C6 Fluorotelomer Foam Loss

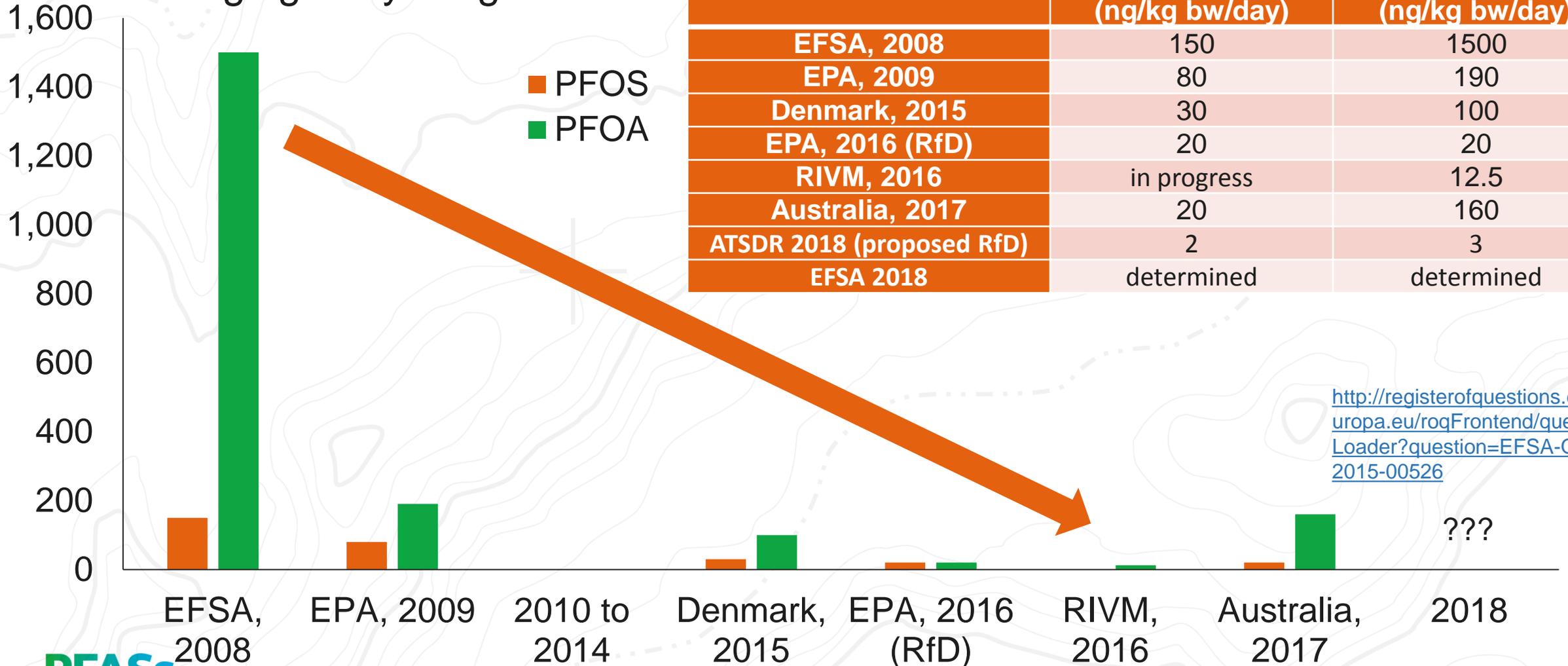


Aerobic Biotransformation Funnel: Conversion of Polyfluorinated Precursors to PFAAs



Tolerable Daily Intake (TDI)

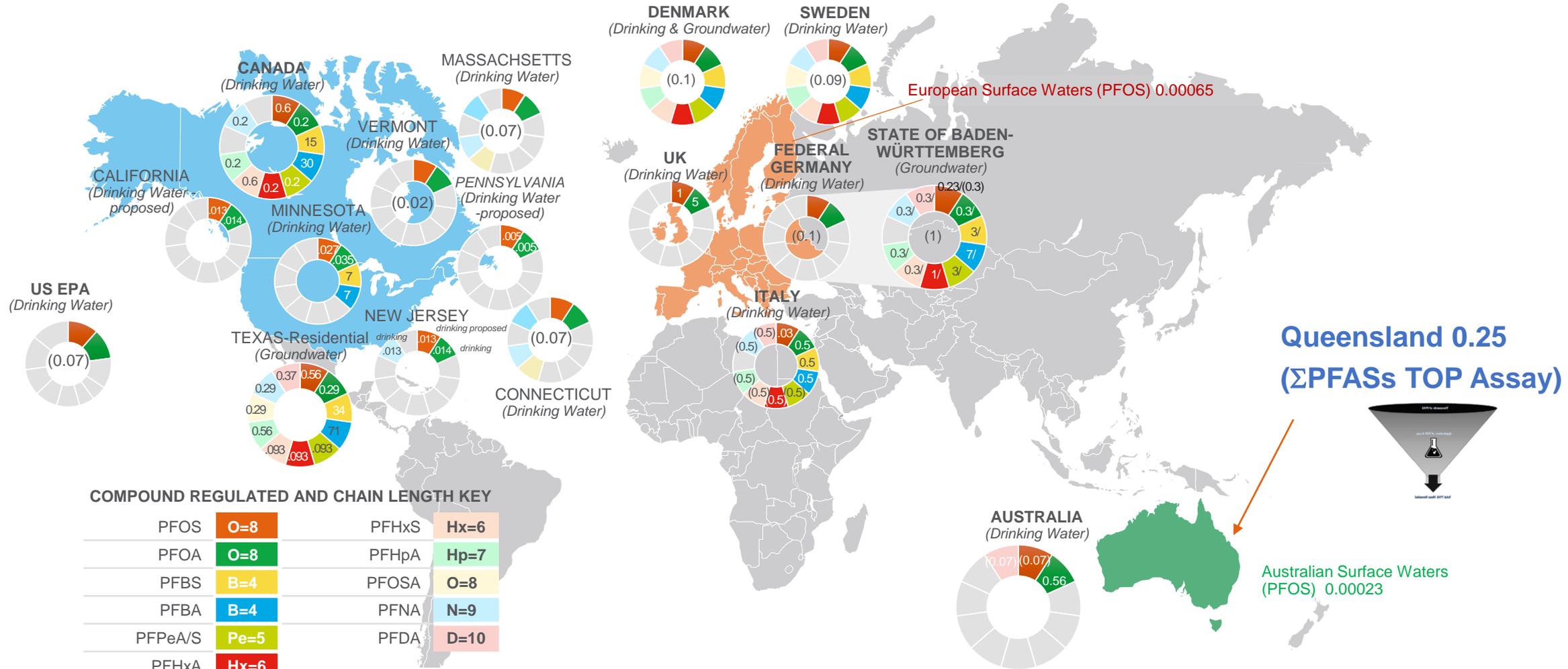
ng/kg/body weight/d



<http://registerofquestions.efsa.europa.eu/roqFrontend/questionLoader?question=EFSA-Q-2015-00526>

???

Evolving Regulatory PFAS Values



Global Regulatory PFAS Tracker

PFAS Global Standards

Region
Asia Australia Europe North America

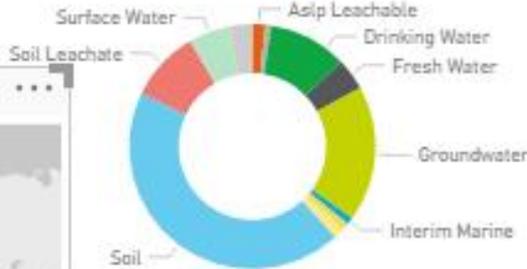
Standards
1765

Proposed or Promulgated

Compound

All

All



Human Exposure to PFASs

Long Chain Human Bioaccumulation Half Life:

PFHxS 8.5 years

PFOS 4.2 years

PFOA 3.8 years

**Drinking Water
And Food**

Main Exposure

House dust

Indoor air

Outdoor air

Consumer products

- Fluoropolymers inc. side chain polymers
- Fluorosurfactants
- Performance chemicals
- Product residuals



Precursor

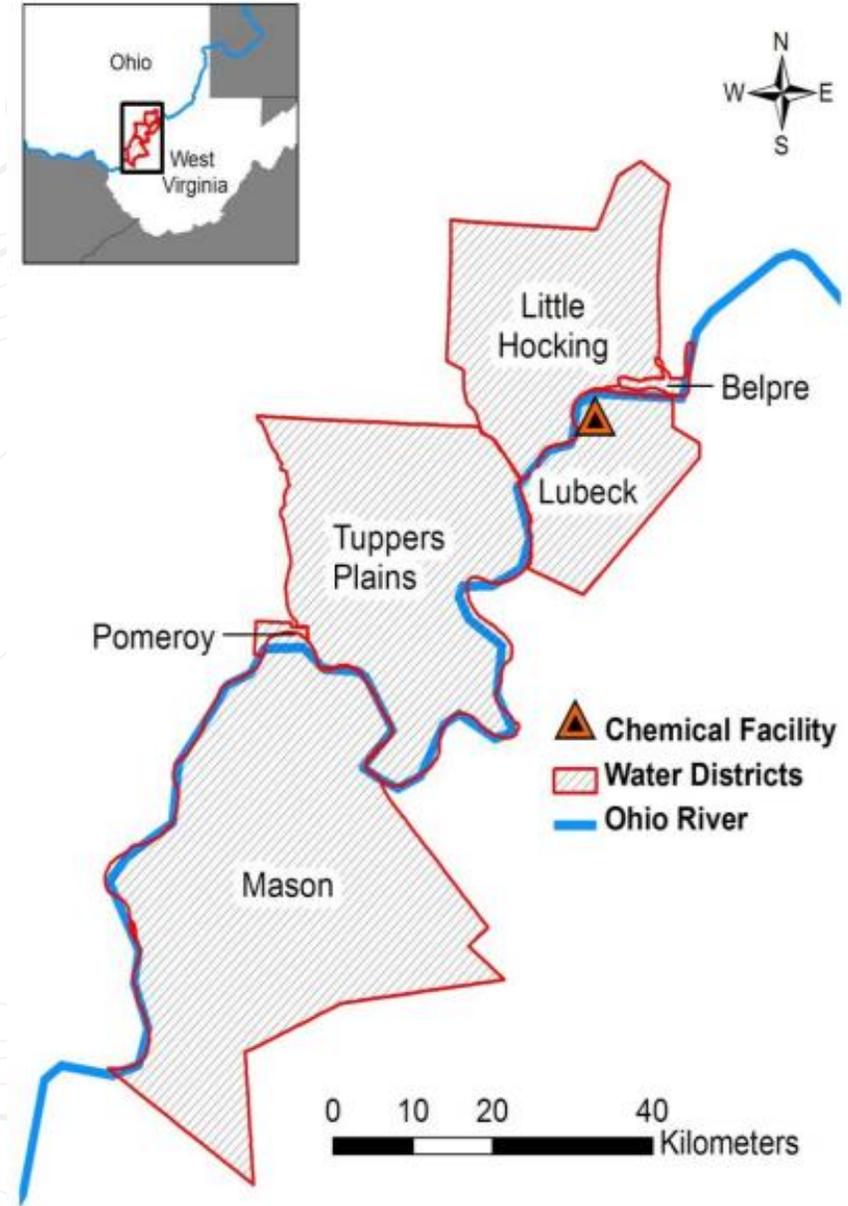
PFAA

C8 Science Panel

- Emissions from manufacturing plant had impacted groundwater used by municipal and private supplies
- Study commissioned by court as part of a class action against manufacturer
- Three epidemiologists studied links between PFOA and various health outcomes
- 55 health outcomes studied, 4 reports issued between 2011 and 2012.
- Information collected (69,030 individuals):
 - Blood biomarkers, 10 PFAA
 - Health questionnaire / medical records
- Study aim was to establish “Given available scientific evidence, is it more probable than not that a connection is present between C8 (PFOA) exposure and disease?”

PFASs

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C8 Science Panel - Findings

- Study found mean PFOA levels of 82.9 ng/L and PFOS levels of 23.6 ng/L in serum samples.
- Probable links were identified for 6 of the 55 health outcomes.

Probable Link



- High Cholesterol
- Ulcerative Colitis
- Thyroid Disease
- Testicular Cancer
- Kidney Cancer
- Pregnancy-Induced Hypertension

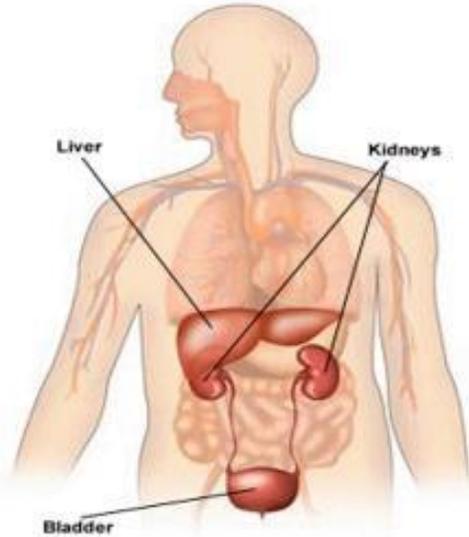
No Probable Link



- Hypertension/Coronary Artery Disease
- Chronic Kidney, Liver, Parkinson's, Autoimmune* Diseases
- Osteoarthritis
- Common Infections
- Neurodevelopmental Disorders/Stroke
- Asthma
- Birth Defects/Preterm/Low Birth Weight
- Miscarriage or Stillbirth

*rheumatoid arthritis, lupus, Type I and II diabetes, Crohn's disease, multiple sclerosis

Toxicity for Humans



- PFAS bind to proteins (not to lipids / fats) and are mainly detected in blood, liver and kidneys
- PFOS: carcinogenicity “suggestive” (US EPA, 2014). PFOA: “possibly carcinogenic” (International Agency for Research on Cancer, IARC, 2014)
- Study with 656 children demonstrated elevated exposure to PFOS & PFOA are associated with reduced humoral immune response ^[1]
- Large epidemiological study of 69,000 persons found probable link between elevated PFOA blood levels and the following diseases: high cholesterol, ulcerative colitis, thyroid disease, testicular cancer, kidney cancer and preeclampsia –C8 science panel ^[2]

[1] Grandjean, P.; Andersen, E. W.; Budtz-Jørgensen, E.; Nielsen, F.; Mølbak, K.; Weihe, P.; Heilmann, C. Serum vaccine antibody concentrations in children exposed to perfluorinated compounds. *JAMA* 2012, 307, 391–397.

[2] <http://www.c8sciencepanel.org/>

Perfluoroalkyl Acid Distribution in Various Plant Compartments of Edible Crops Grown in Biosolids-Amended soils

Andrea C. Blaine,[†] Courtney D. Rich,[†] Erin M. Sedlako,[†] Lakhwinder S. Hundal,[‡] Kuldip Kumar,[‡] Christopher Lau,[§] Marc A. Mills,[#] Kimberly M. Harris,^{||} and Christopher P. Higgins^{†,*}

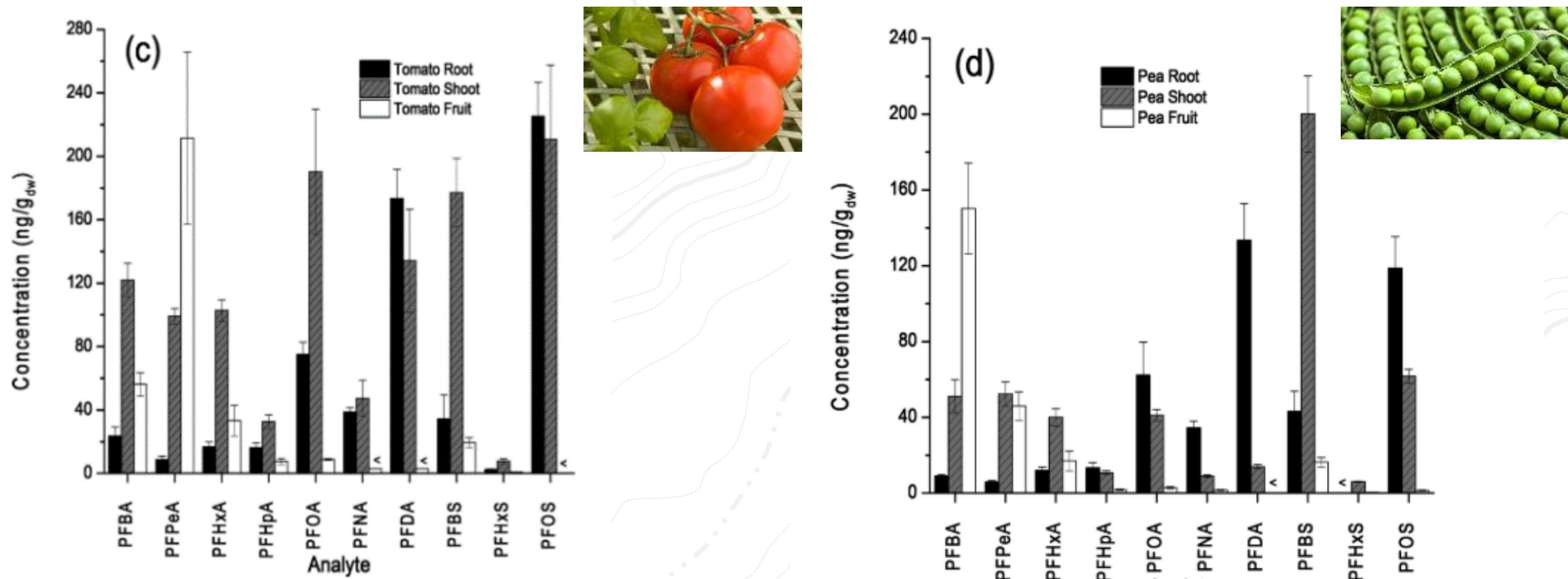


Figure 2. Concentrations of PFAAs in greenhouse radish (a), celery (b), tomato (c), and pea (d) grown in industrially impacted soil. Values for tomato fruit are from a previous study.⁹ Bars represent means and standard errors of five determinations. Values less than the LOQ are denoted by <; LOQs for respective matrix and analyte are listed in SI Table S4 and Table S5.

Concerns over short chain PFASs - Overview

Persistent

- Based on read-across from long chain PFAS
- Long-range transport and findings in remote areas

Mobility and Exposure of Organisms

- Potential to contaminate drinking water resources
- Difficult to be removed from water
- Binding to proteins
- Non-negligible half-lives in organisms
- Enrichment in plants

Toxic

- No indications of ecotoxicity
- Toxicity in humans to be assessed
- Potential endocrine disruptor

- Representatives of European Authorities agreed: Properties are of concern (BA-Workshop in October 2016)
- However, non-classical combination of concerns so far not covered by REACH


 Umwelt
 Bundesamt

https://reach-info.de/dokument/e/short-chain_workshop_summary.pdf

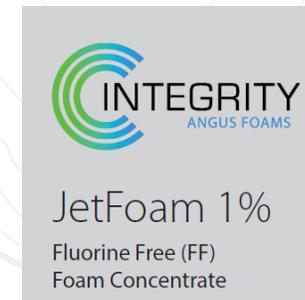
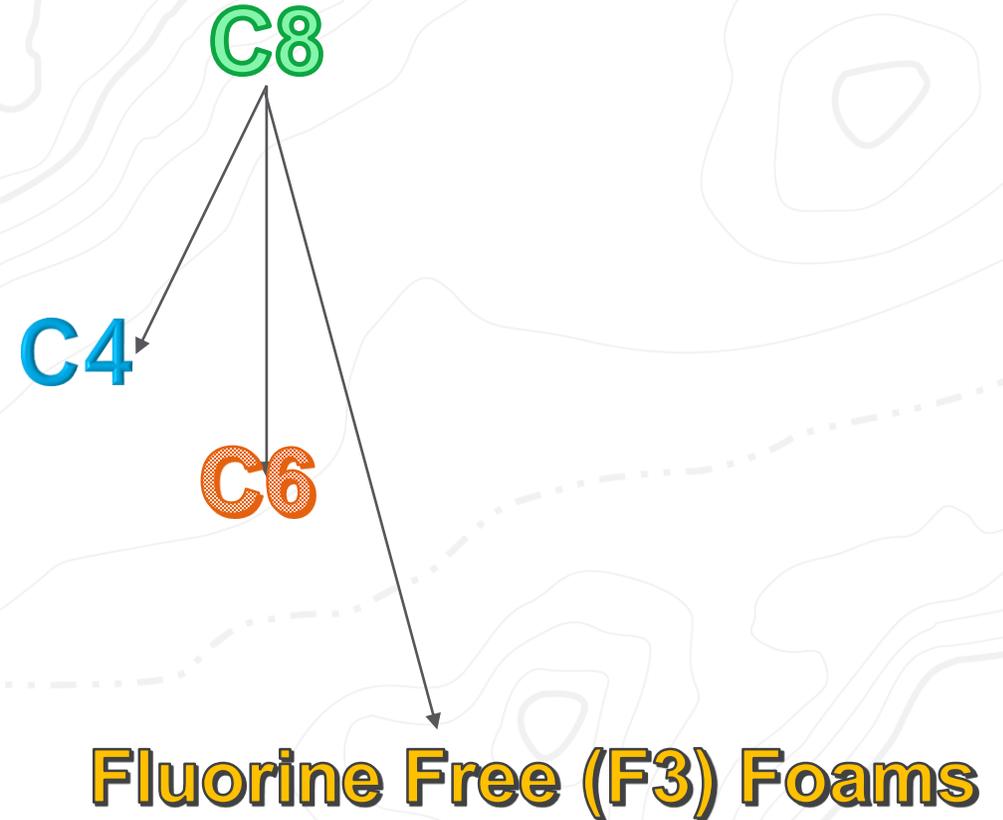
ICCE 2017 Oslo

**Regulation needs support from research:
Short-chain PFASs under REACH**

Lena Vierke, Claudia Staude, Éva Fetter, Stephan Brendel, Annegret Biegel-Engler
 Section IV 2.3 – Chemicals
 German Environment Agency (UBA), Germany

PFAS Foams being Replaced

- C8 (PFOS and PFOA) phased-out
- C8 replaced with compounds with shorter (e.g., C4, C6) perfluorinated chains
- C4, C6 PFAS are less bioaccumulative, still extremely persistent and more mobile in aquifer systems vs C8 - more difficult and expensive to treat in water.
- Solutions for characterizing all PFAS species important to cover current and future risks / liabilities
- Regulations addressing multiple chain length PFAS (long and short) are evolving globally
- Fluorine free (F3) foams contain no persistent pollutants
- F3 foams pass ICAO tests with highest ratings for extinguishment times and burn-back resistance, so are widely available as replacements to AFFF



PFASs in Landfill Leachate

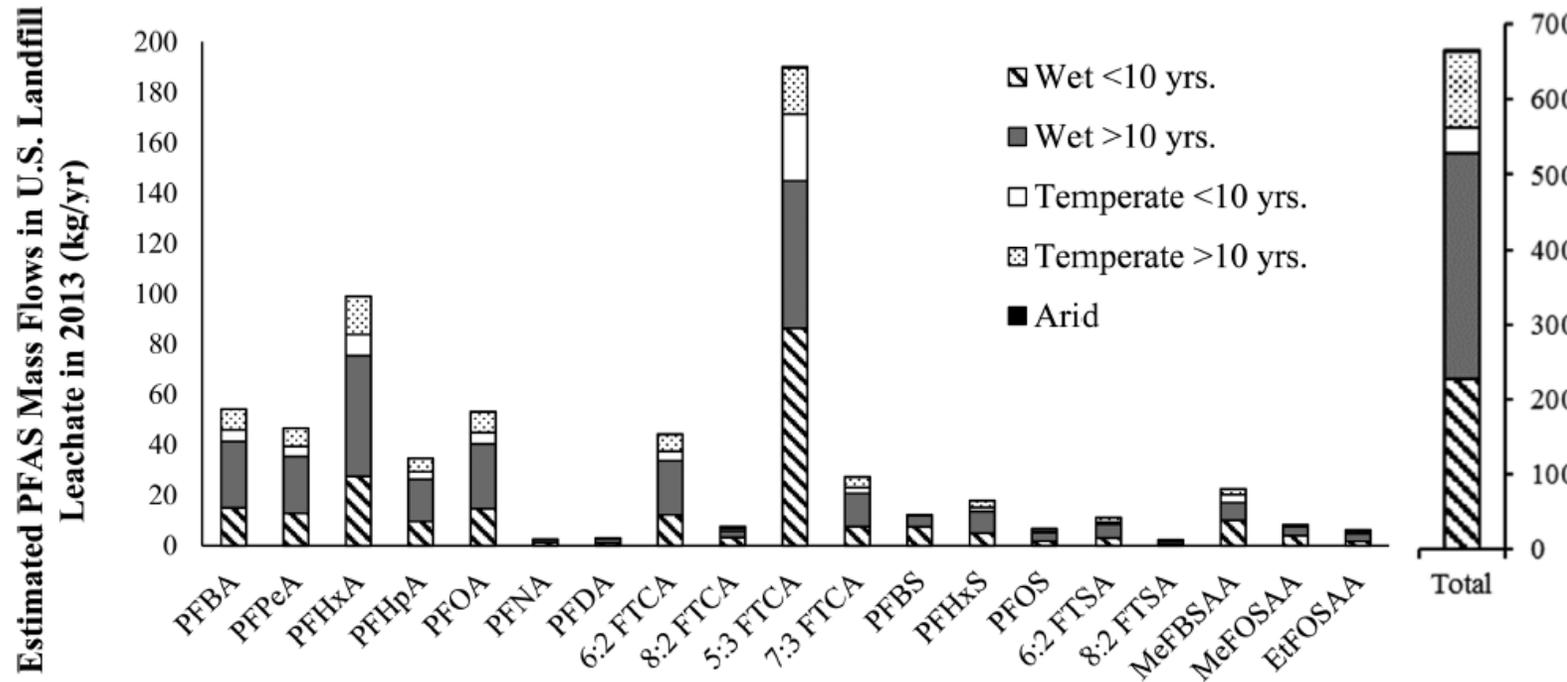


Figure 1 from Lang et.al., (2017) National Estimate of Per- and Polyfluoroalkyl Substance (PFAS) Release to U.S. Municipal Landfill Leachate. *Environ. Sci. Technol.*, 2017, 51 (4), pp 2197–2205

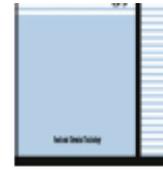
Figure 1. Group 1 PFAS release in U.S landfill leachate for 2013 demonstrating a dominance of compounds with five fluorinated carbons (PFHxA and 5:3 FTCA). Releases were calculated from mean concentrations in each climate and age category (Table 3). The individual columns are based on eq 1 while the total is based on eq 2.

Concerns over short chain PFAS (cont.)



Food and Chemical Toxicology

journal homepage: www.elsevier.com/locate/foodchemtox



ChemicalWatch
GLOBAL RISK & REGULATION NEWS

Short communication

Internal exposure-based pharmacokinetic evaluation of potential for biopersistence of 6:2 fluorotelomer alcohol (FTOH) and its metabolites

Shruti V. Kabadi^{a,*}, Jeffrey Fisher^b, Jason Aungst^a, Penelope Rice^a

^a FDA/CFSAN/OFAS/DRCN, 5001 Campus Drive, HFS 275, College Park, MD 20740, United States

^b FDA/NCTR, 3900 NCTR Road, Jefferson, AR 72079, United States



useful for biomonitoring purposes and toxicological evaluation. More importantly, we determined that 5:3 A is an important biomarker for assessment of long-term exposure to 6:2 FTOH as 5:3 A had the highest internal exposure and slowest clearance across species. Furthermore, we concluded that 5:3 A has the potential to reach steady state upon repeated exposure to 6:2 FTOH as its clearance was determined to reduce with increasing 6:2 FTOH exposure. We also identified specific

FDA scientists voice concerns over metabolites of food contact substance

Focus on metabolites of 6:2 fluorotelomer alcohol

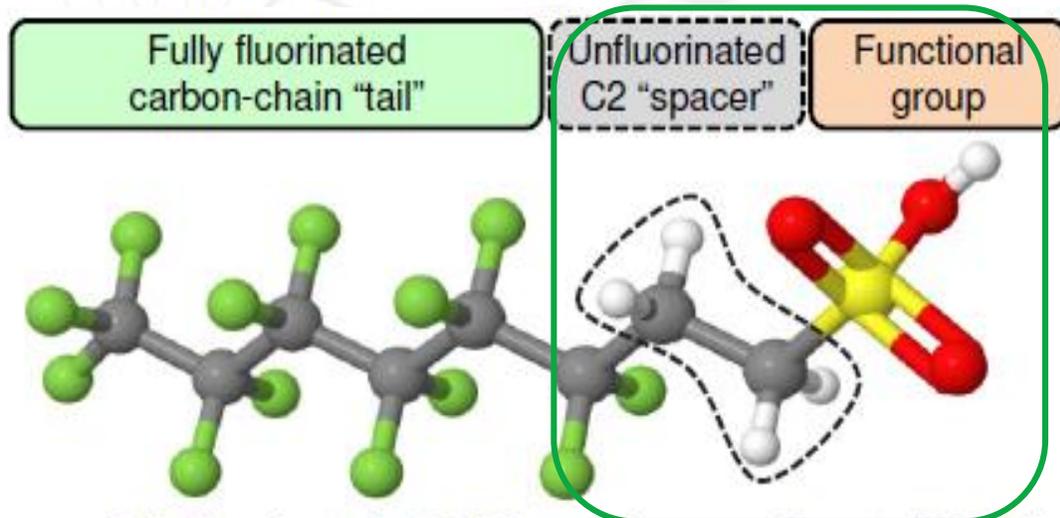
18 January 2018 / Academic studies, Exposure monitoring & measurement, Food & drink, PFCs, Toxicology, United States

"Our work represents the first step towards identifying the mechanism by which 6:2 FTOH, similar C6-PFCs, and its metabolites could accumulate in the body to potentially cause adverse effects," write the researchers in the journal *Food and Chemical Toxicology*.

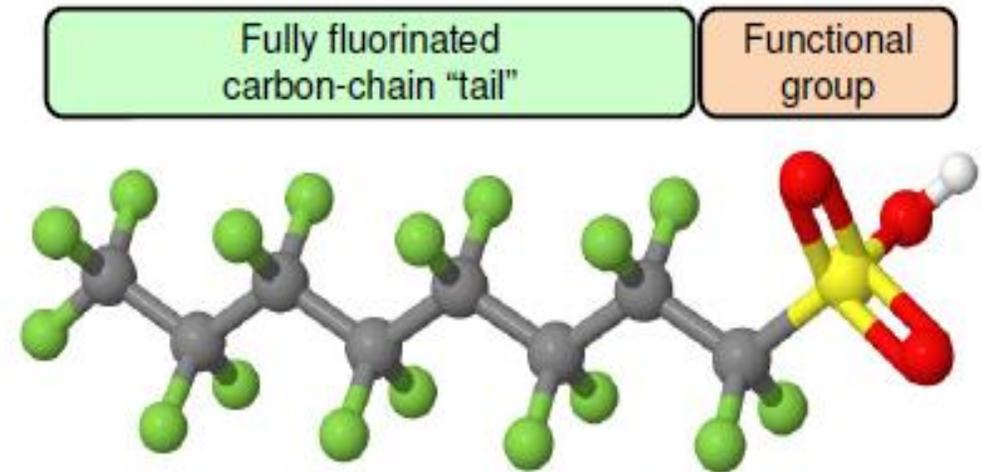
PFASs

© Arcadis 2016

Polyfluorinated Compounds – PFAA Precursors



e.g. Poly-fluorinated - 6:2 Fluorotelomer sulfonate (6:2FtS)



e.g. Per-fluorinated - PFOS

Atoms—Green=Fluorine, Grey=Carbon, White=Hydrogen, Red=Oxygen, Yellow=Sulphur, non-fluorinated carbon atoms circled

Bioactive –metabolised via reactive aldehydes and creates secondary molecule—increased toxicity vs inert PFAA



Estimating human exposure to PFOS isomers and PFCA homologues: The relative importance of direct and indirect (precursor) exposure

Wouter A. Gebbink*, Urs Berger, Ian T. Cousins

162

Department of Applied Environmental Science (ITM), Stockholm University, SE 10691 Stockholm, Sweden

W.A. Gebbink et al. / Environment International 74 (2015) 160–169

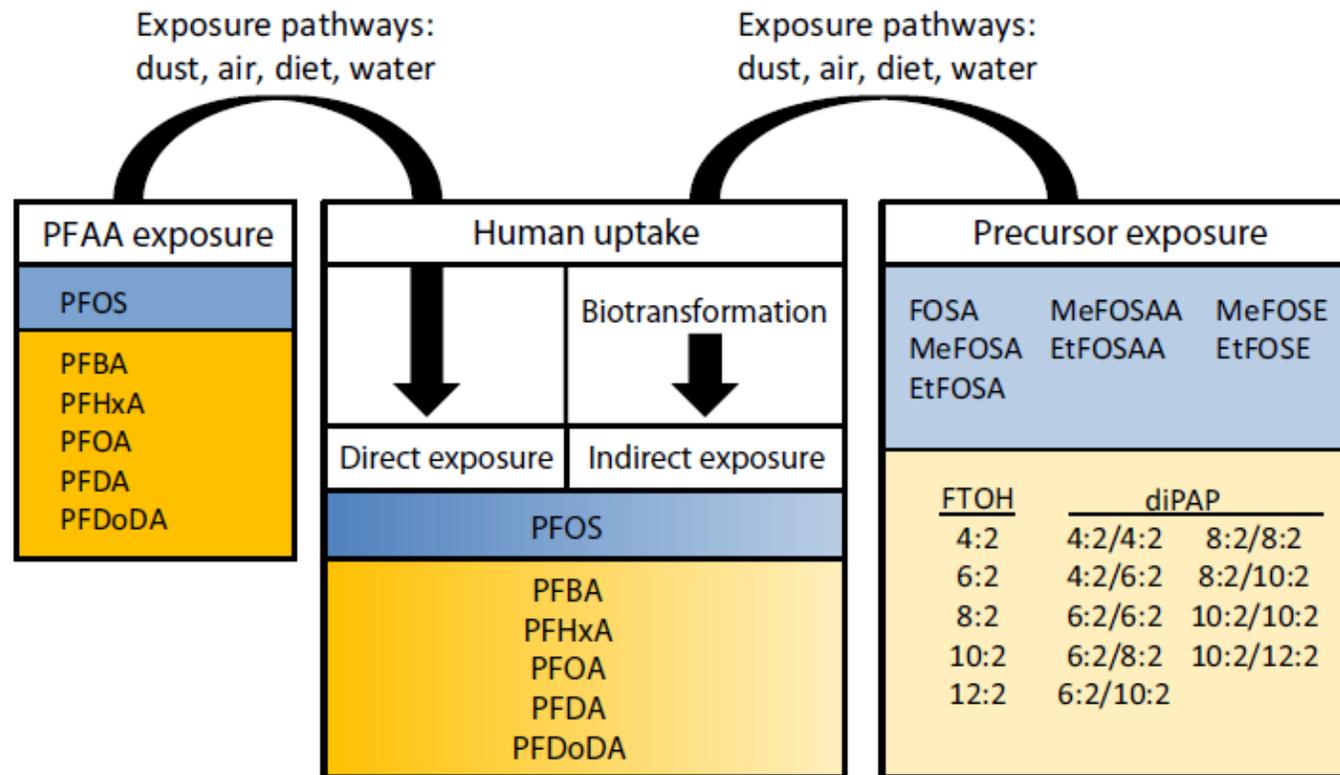
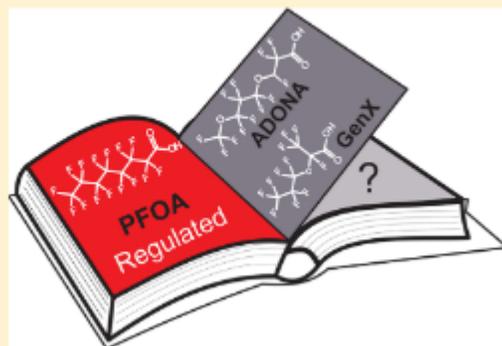


Fig. 1. Schematic of direct and indirect (precursor) exposure pathways for PFOS and PFCAs.

A Never-Ending Story of Per- and Polyfluoroalkyl Substances (PFASs)?

Zhanyun Wang,[†] Jamie C. DeWitt,[‡] Christopher P. Higgins,[§] and Ian T. Cousins^{*,||}

ABSTRACT: More than 3000 per- and polyfluoroalkyl substances (PFASs) are, or have been, on the global market, yet most research and regulation continues to focus on a limited selection of rather well-known long-chain PFASs, particularly perfluorooctanesulfonate (PFOS), perfluorooctanoic acid (PFOA) and their precursors. Continuing to overlook the vast majority of other PFASs is a major concern for society. We provide recommendations for how to proceed with research and cooperation to tackle the vast number of PFASs on the market and in the environment.



Fluorinated alternatives to long-chain perfluoroalkyl carboxylic acids (PFCAs), perfluoroalkane sulfonic acids (PFSA)s and their potential precursors

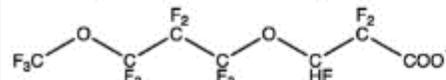
Zhanyun Wang^a, Ian T. Cousins^b, Martin Scheringer^{a,*}, Konrad Hungerbühler^a

^a Institute for Chemical and Bioengineering, ETH Zurich, Wolfgang-Pauli-Strasse 10, CH-8093 Zurich, Switzerland

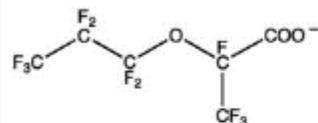
^b Department of Applied Environmental Science (ITM), Stockholm University, SE-10691 Stockholm, Sweden

Fluoropolymer manufacture

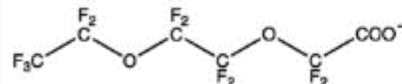
ADONA (CAS No. 958445-44-8)



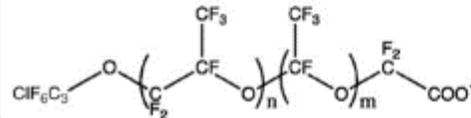
GenX (CAS No. 62037-80-3)



Asahi's product (CAS No. 908020-52-0)

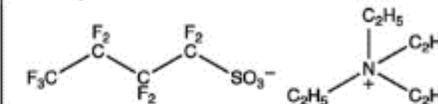


Solvay's product (CAS No. 329238-24-6)

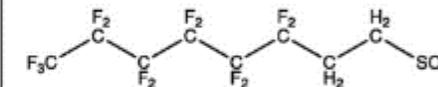


Metal plating

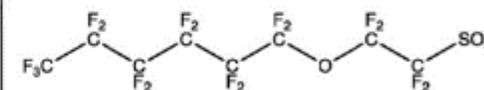
N(Et)₄-PFBS (CAS No. 25628-08-4)



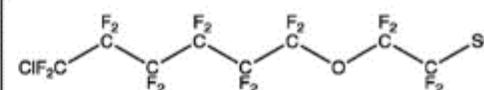
6:2 FTSA (CAS No. 27619-97-2)

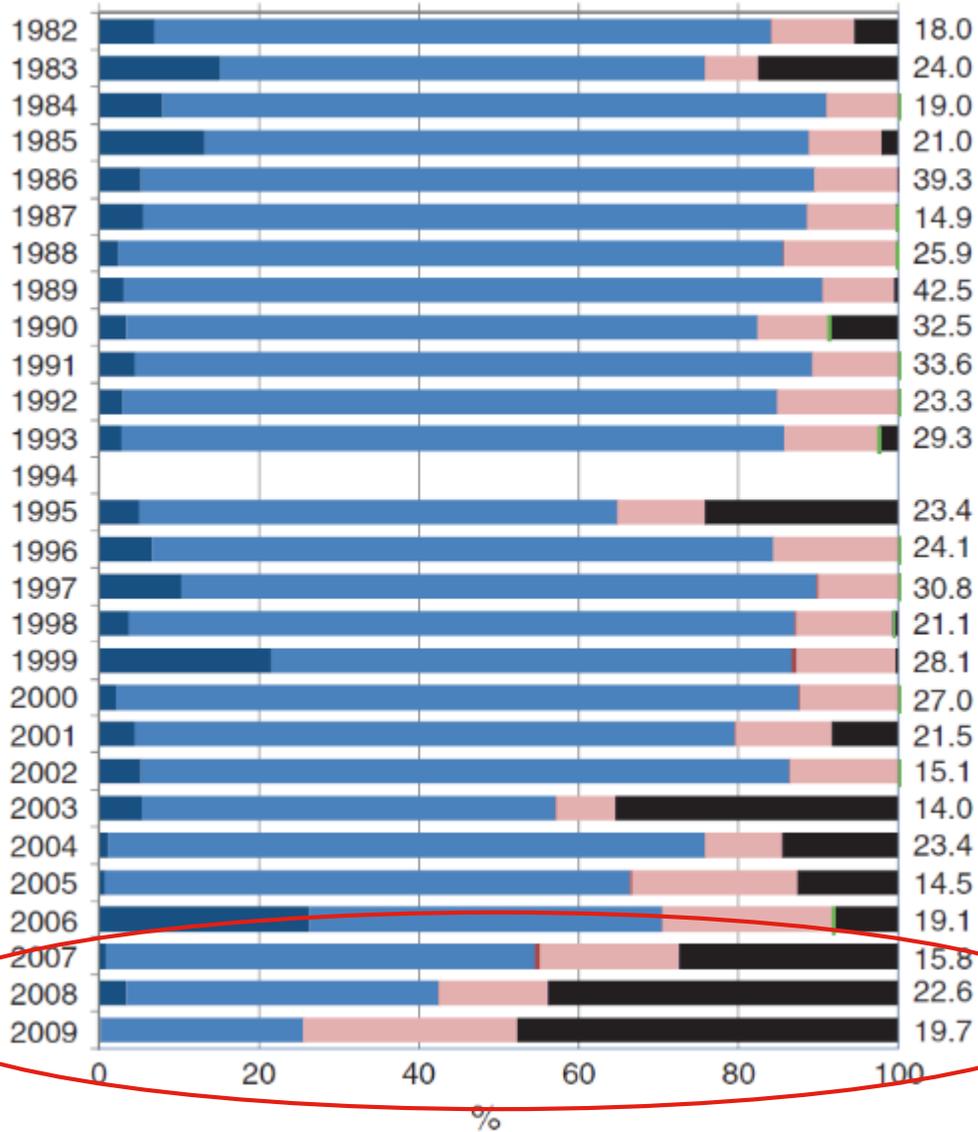


F-53 (CAS No. 754925-54-7)



F-53B (CAS No. 73606-19-6)





Are humans exposed to increasing amounts of unidentified organofluorine?

Leo W. Y. Yeung^A and Scott A. Mabury^{A,B}

Plasma concentrations reveal transition from identifiable PFASs to unidentified

Criteria to Categorize Safety of Chemicals

- Persistence (**P**): Compounds that do not break down in the environment over long periods of time (i.e., they do not readily biodegrade).
- Bioaccumulative (**B**): Compounds that build up and are retained in organisms at a faster rate than they can be removed or expelled.
- Mobility (**M**): Compounds that can travel long distances in groundwater or surface waters from their point of release.
- Toxicity (**T**): Compounds impart an adverse health effect to an organism at a relatively low concentration of exposure.
- Biopersistence: Compounds that tend to remain inside an organism, rather than being expelled or broken down.
- Biomagnification: The increased concentration of a compound, such as a toxic chemical, in the tissues of organisms at successively higher levels in the food chain –through trophic levels.

Stockholm Convention on Persistent criteria to define persistent organic pollutants (POPs), based on **PBT**

Suggestion that other criteria are more relevant by UBA under REACH: **PMT, vPvB, vPvM**

The Precautionary Principle

Principle 15 of the Rio Declaration on Environment and Development as follows:

“Where there are threats of serious or irreversible damage, lack of full scientific certainty shall not be used as a reason for postponing cost-effective measures to prevent environmental degradation”

Comparisons are made to the medical Hippocratic Oath – “Do No Harm”

Multiple ways to apply this concept: Persistent Bioaccumulative and Toxic (PBT), very persistent and very bioaccumulative (vPvB), very mobile very persistent (vPvM), Persistent Mobile Toxic (PMT) or poorly reversible exposure

https://www.epa.vic.gov.au/~media/Files/Your%20environment/Land%20and%20groundwater/PFAS%20in%20Victoria/PFAS%20NEMP/FINAL_PFAS-NEMP-20180110.pdf

Poor Reversibility of Exposure (Conceptual)

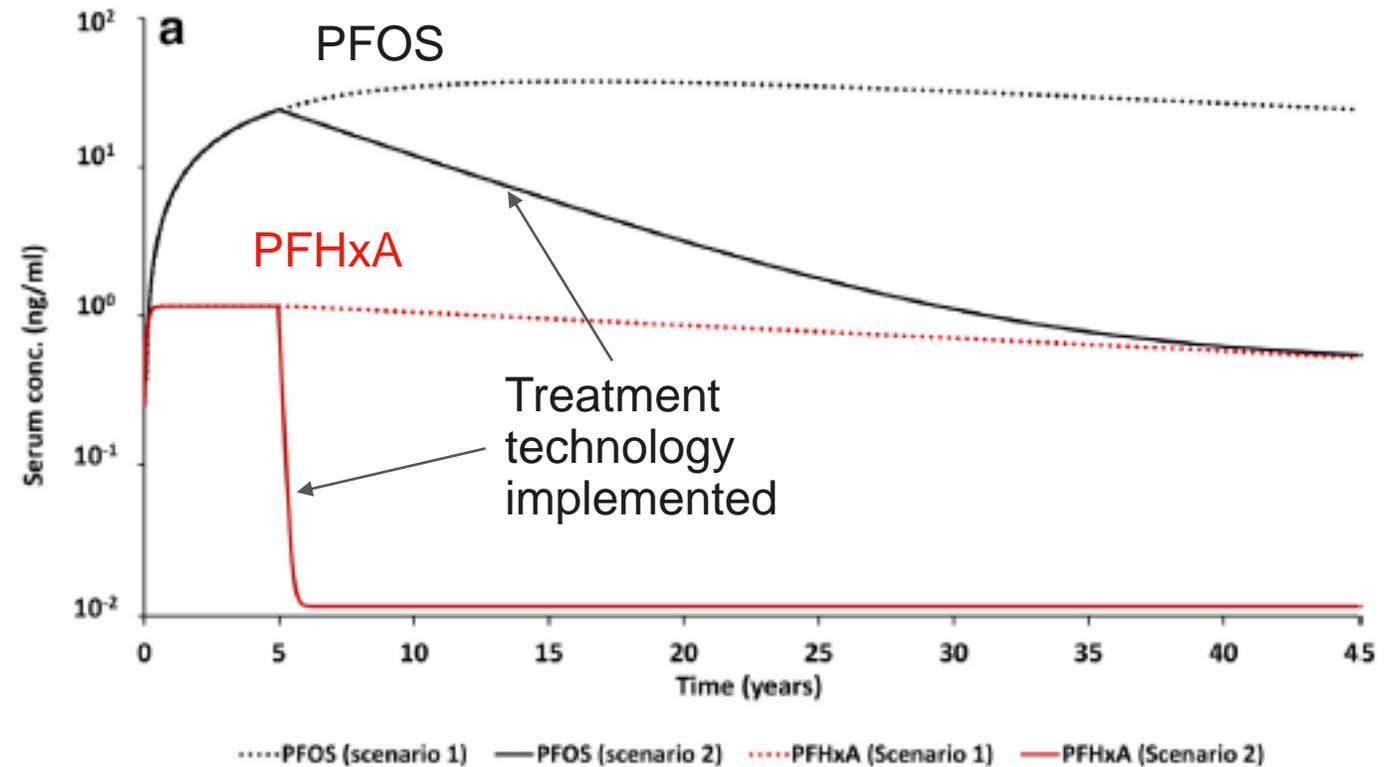
Poorly reversible exposure of a chemical can occur two ways:

- The chemical has slow elimination kinetics in organisms (bioaccumulative)

or

- Due to environmental recalcitrance, exposure is steady (extreme persistence)

(Cousins *et al.* Environ. Int. 2016)



Major Locations of PFAS Point Source Contamination

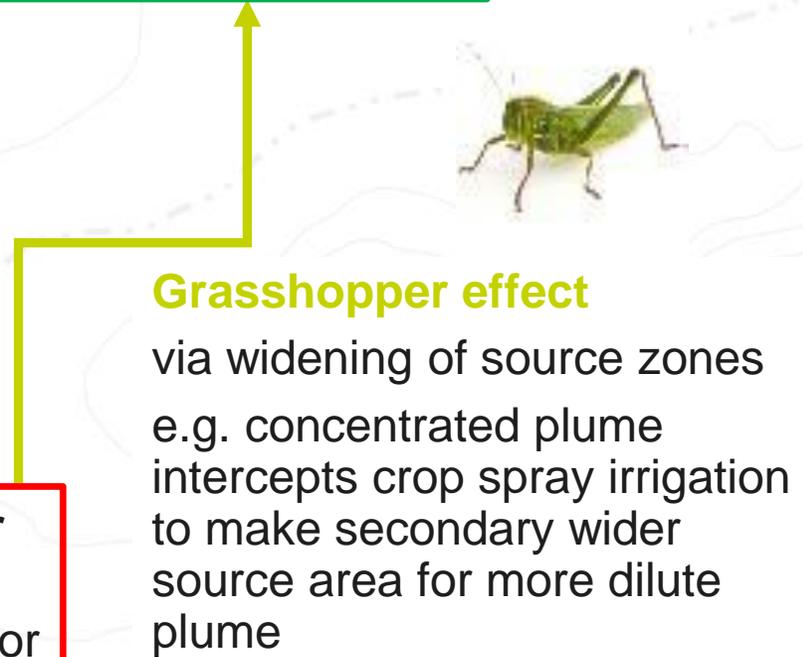
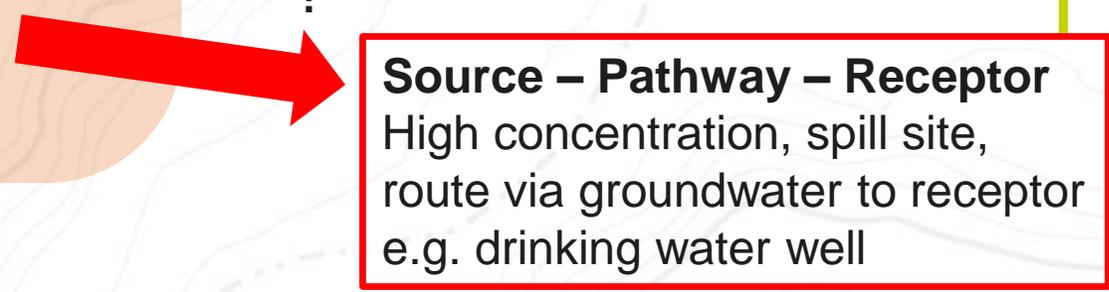
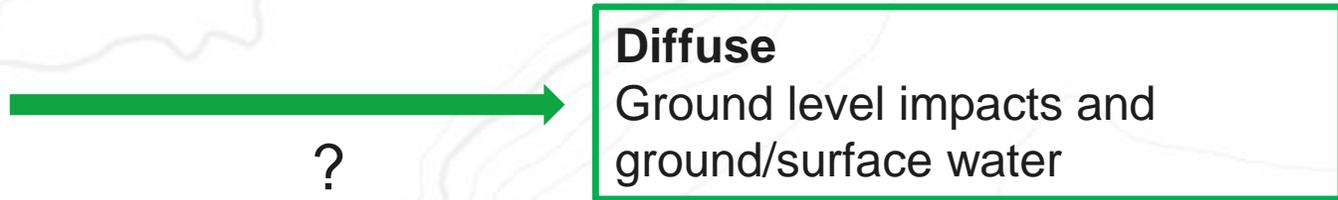
- Primary Manufacturing (e.g. PTFE)
- Secondary Manufacturing (Application of PFASs to other products) –Tanneries, Paper Mills, Firefighting Foam Blending, Metal Plating Facilities
- Fire Training Sites
 - Airports
 - Civil
 - Defence
 - Oil and Gas
 - Large Rail Yards
- Wastewater treatment plants
- Landfills



Groundwater Risks to Receptors

Landfill Leachate
Municipal / Domestic WWTP
Industry & Manufacturing
Agricultural Land
Commercial / Domestic Products
Metal Plating
ASTs –Fuel storage (FFFP / FP)

AFFF / FFFP / FP
Fire training
Incident Response





using science to
create a better place

Incidence and attenuation of perfluorinated surfactants in groundwater

Science Report – SC070002/SR

Perfluorinated chemical	Limit of detection (µg/l)
Perfluorooctane sulphonate (PFOS)	0.1
Perfluoropentanoic acid (PFC5)	0.1
Perfluorohexanoic acid (PFC6)	0.1
Perfluoroheptanoic acid (PFC7)	0.1
Perfluorooctanoic acid (PFOA)	0.1
Perfluorononanoic acid (PFC9)	0.1
Perfluorodecanoic acid (PFC10)	0.1
Perfluoroundecanoic acid (PFC11)	0.1
Perfluorododecanoic acid (PFC12)	0.1
Perfluorotetradecanoic acid (PFC14)	0.1

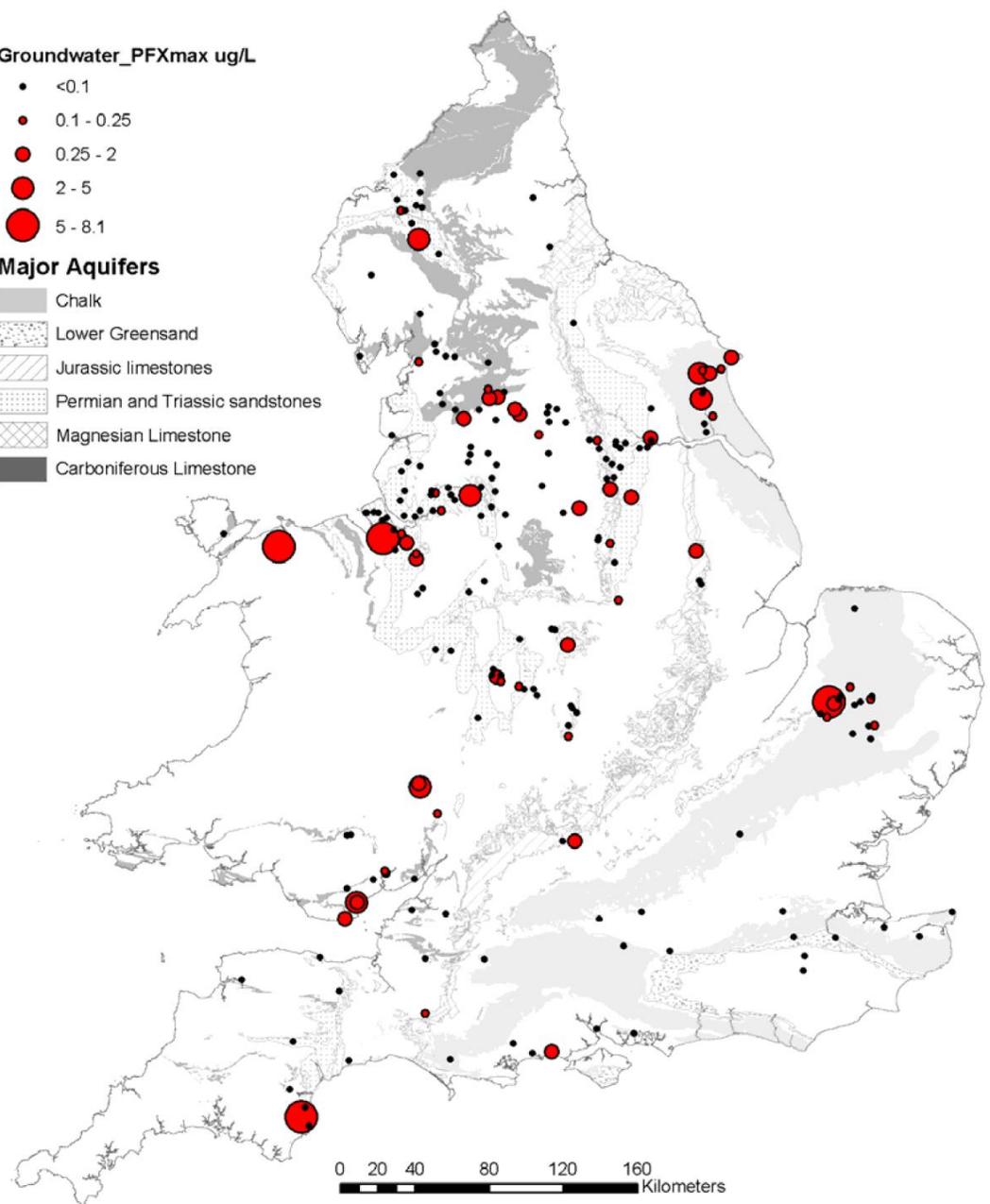
- Environment Agency PFAA monitoring 2008
- Groundwater sampling
 - Conducted at 219 sites in England and Wales (6.5% EA network)
 - The majority of sites were in areas of potential sources eg. airfields
 - “Low risk rural sites” comprised 5%
- Surface water sampling
 - Drinking water abstractions (42 sites)
 - “Higher risk sites” (39 sites) eg. effluents from sewage works
- Limits of detection were 0.1 µg/L so well above the new US EPA standards at 0.07 µg/L

Groundwater_PFXmax ug/L

- <0.1
- 0.1 - 0.25
- 0.25 - 2
- 2 - 5
- 5 - 8.1

Major Aquifers

- Chalk
- Lower Greensand
- Jurassic limestones
- Permian and Triassic sandstones
- Magnesian Limestone
- Carboniferous Limestone



Groundwater

- PFCs detected in 26% of groundwater sites
- Detection even at “low risk” sites

Surface Water

- PFCs detected at 52% of surface water sites (drinking water abstractions)
- PFCs detected at 67% “high risk” sites

Table 3.1 Perfluorinated compounds in groundwater by aquifer type

Aquifer	Number of sites monitored	Sites with detected PFX		Max PFX (ug L ⁻¹) excluding non detections			
		Number of sites	% sites	Min	Mean	Median	Max
Drift	6	1	16.7	0.2	0.20	0.20	0.2
Minor	75	18	24.0	0.12	1.18	0.39	6.56
Chalk	36	13	36.1	0.1	1.35	0.22	8.1
L'wer GS	3		0.0				
Jur's Lst	3	2	66.7	0.22	1.04	1.04	1.85
PT sst	72	14	19.4	0.1	1.46	0.31	7.47
Mag Lst	7	3	42.9	0.1	0.30	0.20	0.6
Carb Lst	16	6	37.5	0.12	0.39	0.45	0.64

Figure 3.1 Distribution of groundwater monitoring for perfluorinated chemicals

PFAS in European Surface Waters

River	PFOS (ng/l)	Flow(m ³ /s)
Scheldt (Be, NL)	154	-
Seine (Fr)	97	80
Severn (UK)	238	33
Rhine (Ge)	32	1,170
Krka (SI)	1,371	50



EU-wide survey of polar organic persistent pollutants in European river waters

Robert Loos*, Bernd Manfred Gawlik, Giovanni Locoro, Erika Rimaviciute, Serafino Contini, Giovanni Bidoglio

European Commission, Joint Research Centre, Institute for Environment and Sustainability, Via Enea 189, 21020 Spres, Italy

More than 100 river water samples from 27 European Countries were analysed for 35 selected polar organic contaminants.

Table 5.2: EQS of the European Commission for PFOS and its derivatives

Name of substance	AA-EQS* (µg/l)		MAC-EQS** (µg/l)		EQS (µg/kg)
	Inland surface waters	Other surface waters	Inland surface waters	Other surface waters	Biota
Perfluoro octane sulfonate and its derivatives (PFOS)	0,00065	0,00013	36	7,2	9,1

* AA: Annual average

** MAC: Maximum allowable concentration

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Worldwide Distribution of Novel Perfluoroether Carboxylic and Sulfonic Acids in Surface Water

Yitao Pan,^{†,‡,⊥} Hongxia Zhang,^{†,⊥} Qianqian Cui,[‡] Nan Sheng,[‡] Leo W. Y. Yeung,[§] Yan Sun,^{||} Yong Guo,^{||} and Jiayin Dai^{*,†,⊕}

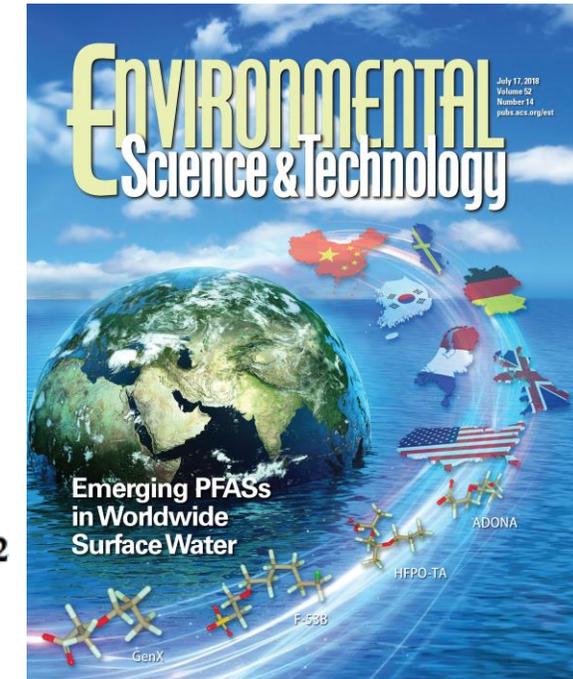
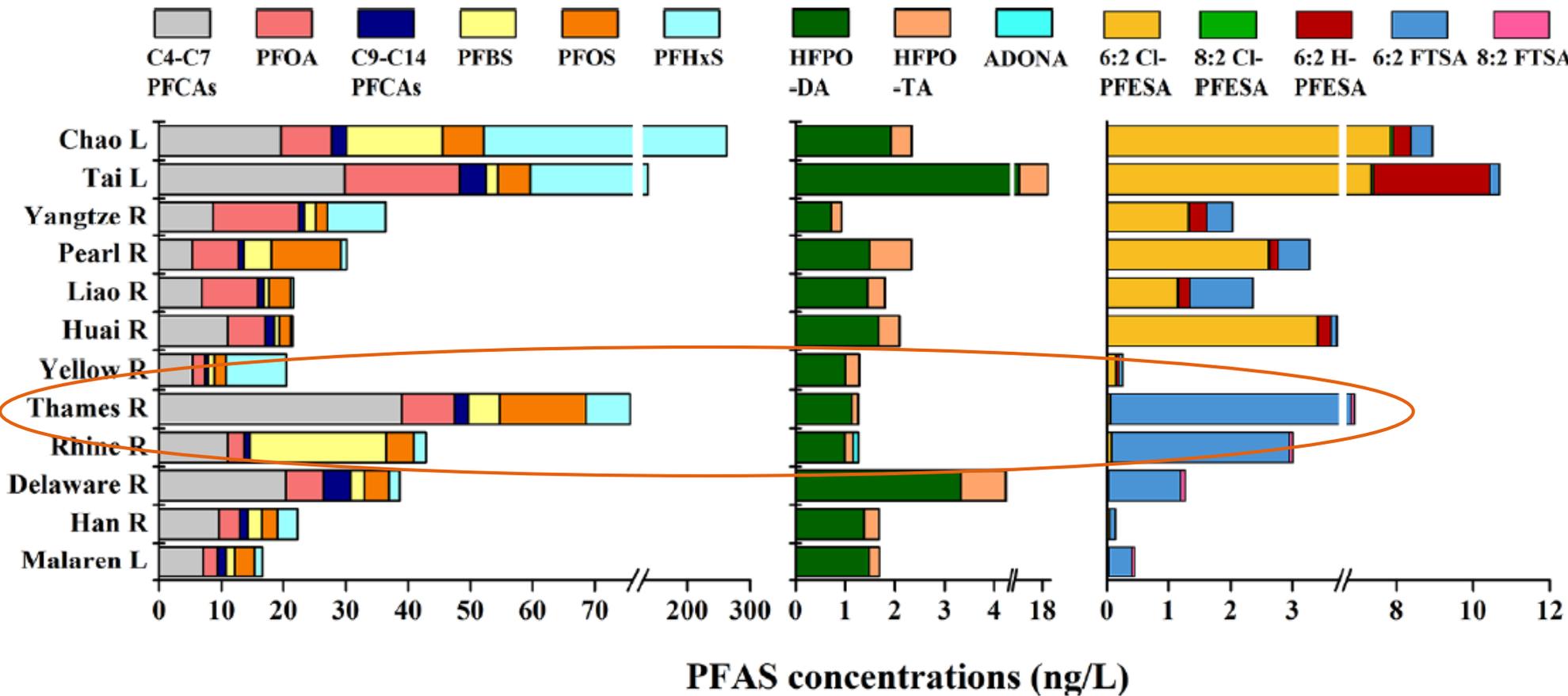


Figure 2. Mean concentrations (ng/L) of legacy PFASs (PFCAs and PFSAs) and fluorinated alternatives (PFECAs, PFESAs, and FTSAs) in the studied rivers and lakes: Chao Lake ($n = 13$), Tai Lake ($n = 15$), Yangtze River ($n = 35$), Pearl River ($n = 13$), Liao River ($n = 6$), Huai River ($n = 9$), Yellow River ($n = 15$), Thames River ($n = 6$), Rhine River ($n = 20$), Delaware River ($n = 12$), Han River ($n = 6$), and Mälaren Lake ($n = 10$).

Occurrence of PFCs in UK Drinking Water



SURVEY OF THE PREVALENCE OF PERFLUOROOCTANE SULPHONATE (PFOS), PERFLUOROOCTANOIC ACID (PFOA) AND RELATED COMPOUNDS IN DRINKING WATER AND THEIR SOURCES

WRc Ref: DEFRA 7585
FEBRUARY 2008



- Sampling and a data review was conducted by the Drinking Water Inspectorate in 2007
- Review found only 4 of 29 water companies had PFAS testing data to share with the DWI
- The sampling drinking water at raw, mid treatment, and final stage of process
 - Fifteen “high risk” sites based on surrounding land use (usually airfields)
 - Five lower risk “control” sites

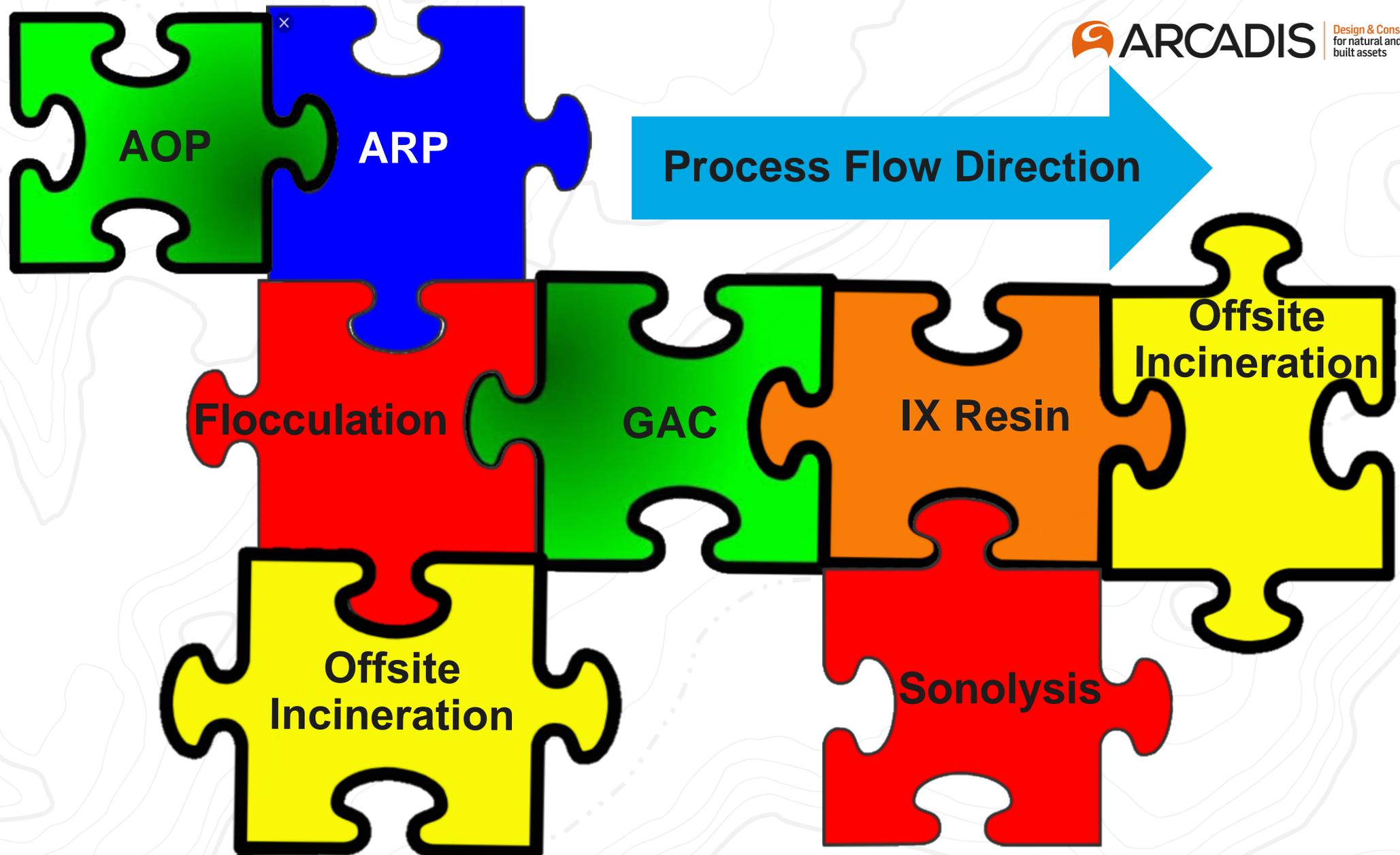
http://dwi.defra.gov.uk/research/completed-research/reports/DWI70_2_212PFOS.pdf

Table 6.2 PFOS results (in µg/l) of sampling in 2007 at the perceived higher-risk sites for all four sampling sessions.

Site ID	Groundwater (UC)	Treatment	Source	Raw	0.016	0.020	<0.011	0.013	for natural and built assets
				Before last chlorination	0.016	0.023	<0.011	-	
				After Cl contact tanks	0.016	0.019	<0.011	0.014	
8	Groundwater (UC)	Ultrafiltration membranes and chlorination using chlorine gas.	Airfield and industrial	Raw	0.016	0.020	<0.011	0.013	
				Before last chlorination	0.016	0.023	<0.011	-	
				After Cl contact tanks	0.016	0.019	<0.011	0.014	
				Final	-	-	-	0.130	
9	Groundwater (UC)	None on site: water pumped by raw water pumping main to another treatment works, which has GAC (not in operation), super chlorination using gas and dechlorination.	Airfield	Raw	0.152	0.124	0.205	0.135	
				Raw	0.162	-	0.183	-	
				Raw	0.154	-	0.208	-	
				Final	-	-	-	0.130	
15	Groundwater (UC)	GAC and chlorination by on-site electrolytic generation using food grade salt stored in HDPE. 12 GAC beds.	Large use of PFOS-containing fire fighting foam	Borehole 3	0.059	0.076	0.052	Off line	
				Borehole 4	0.029	0.028	0.018	Off line	
				Borehole 5	0.038	0.029	0.030	Off line	
				GAC feed water	-	-	-	0.046	
				GAC 1	-	-	-	0.042	
				GAC 2	-	-	-	0.043	
				GAC 3	-	-	-	0.044	
				GAC 4	-	-	-	0.048	
				GAC 5	-	-	-	0.046	
				GAC 6	-	-	-	0.044	
				GAC 7	-	-	-	0.046	
				GAC 8	-	-	-	0.044	
				GAC 9	-	-	-	0.038	
				GAC 10	-	-	-	0.038	
				GAC 11	-	-	-	0.039	
				GAC 12	-	-	-	0.037	
				Post GAC (all)				0.042	
After chlorine contact				0.045	0.040	0.032	0.034		

Available *In Situ* Groundwater Treatment Technologies

Technology	Likelihood of Success?	Rationale
Aerobic Biodegradation	Low	Biotransformation does not proceed past PFAA
Anaerobic Biodegradation	Low	
Phytoremediation	Low	
Air Sparging/Vapor Extraction	Low	
In-Situ Thermal Treatment	Low	
Chemical Oxidation/Reduction	Moderate	Bench-tests confirm viable mechanisms exist; field evidence conflicting; potential for treatment train
Monitored Natural Attenuation	Moderate	PFAA do not biodegrade (dilution via advection, adsorption are viable mechanisms)
Groundwater Extraction and Ex-Situ Treatment	High	Presumptive remedy for PFAS to-date, focus of this discussion; ex-situ waste management
Permeable Reactive Barriers	High	Apply ex-situ sorption technologies with a funnel & gate; change outs required



C6 Firefighting Foam Loss

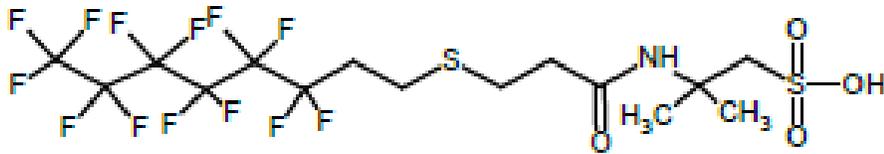
KEMI
Swedish Chemicals Agency

Chemical Analysis of Selected Fire-fighting Foams on the Swedish Market 2014

PM 6/15



Tentatively identified PFAS as a main ingredient is 6:2 FTSAS (fluorotelomermercaptoalkylamido sulfonate).

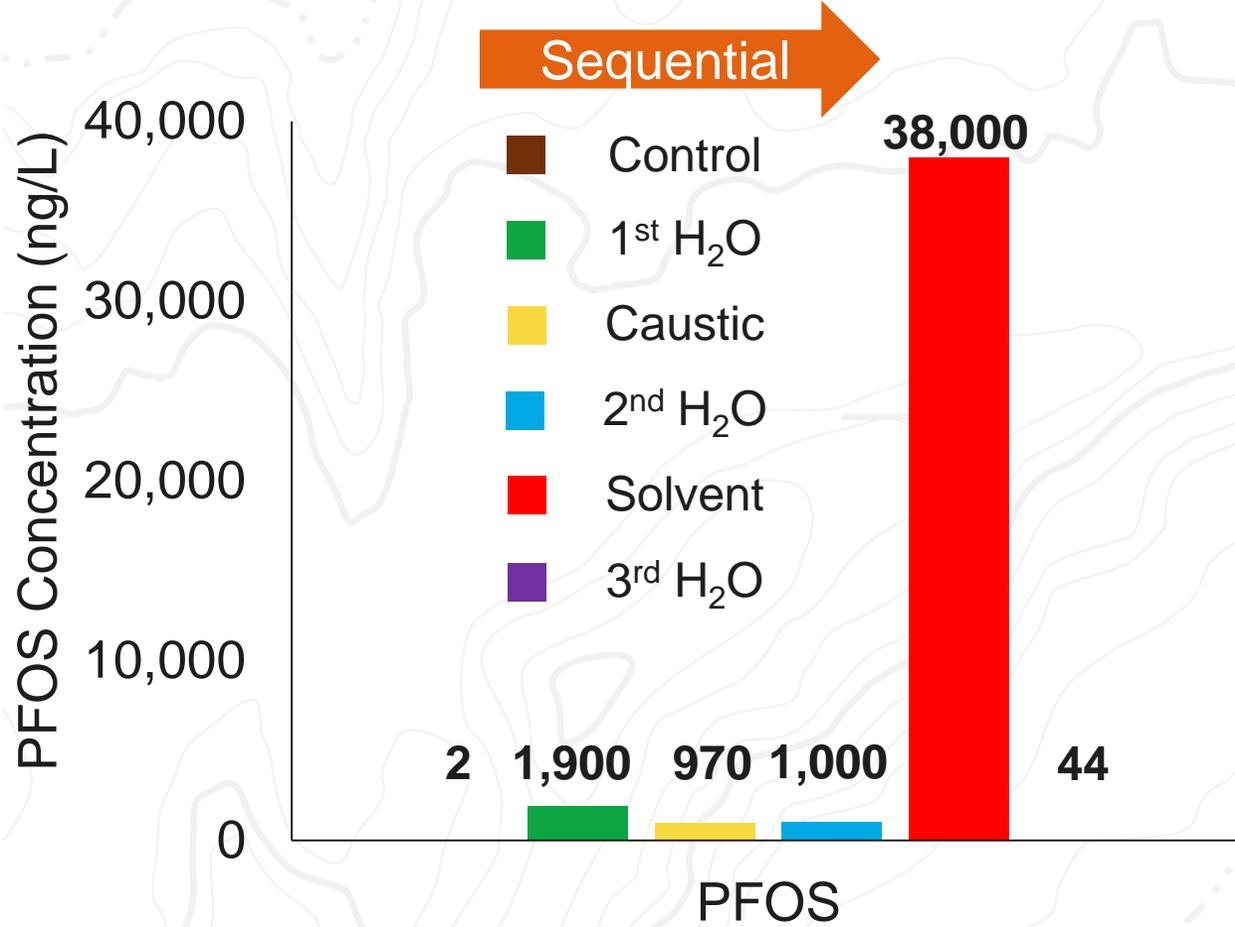


PFASs

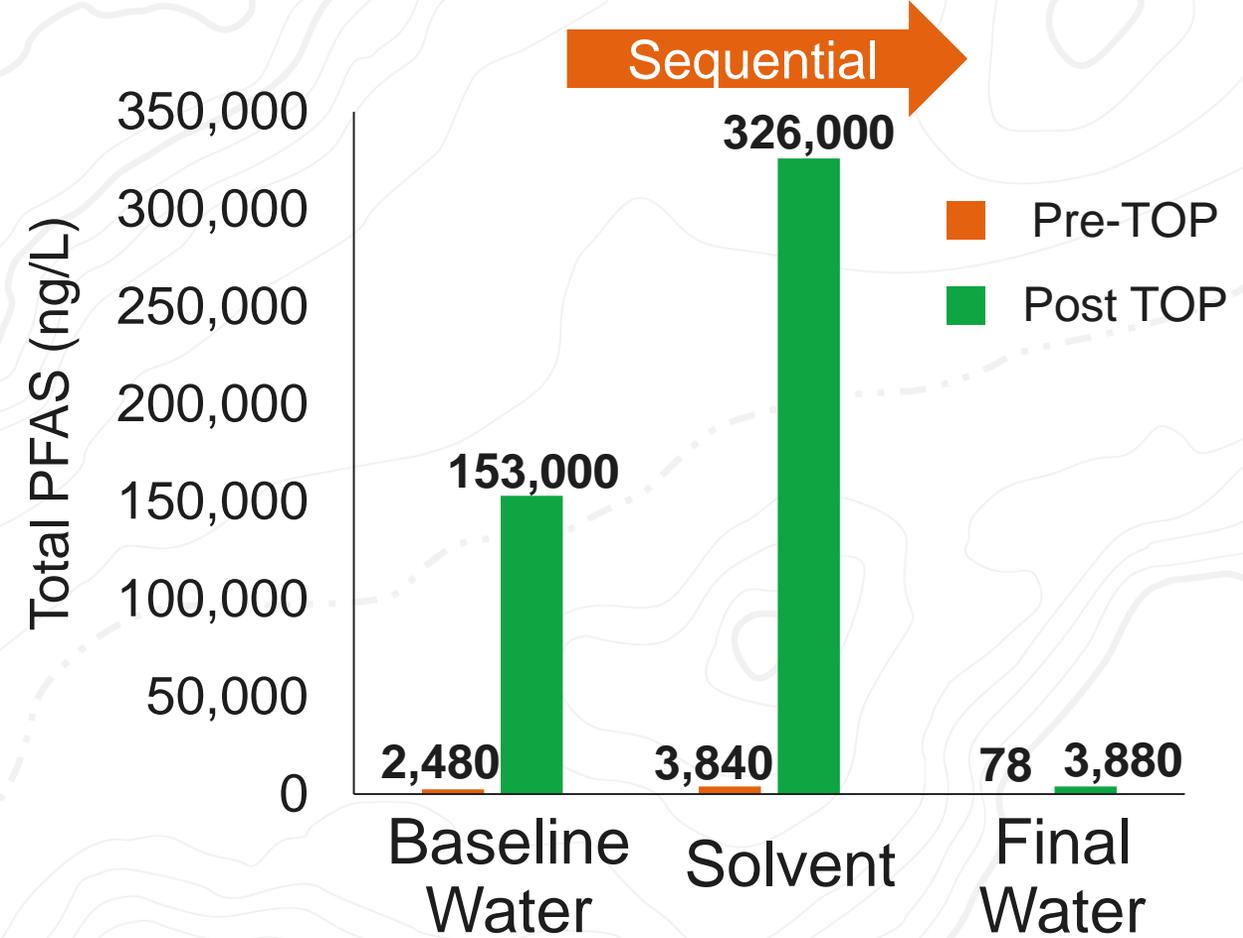
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Foam Cleanout/Decontamination

Sewer Decontamination Trial



Foam Tank Cleanout – TOP Assay



The Challenge

- **+ 15 ML** - Sewage, Tradewaste, Brackish Creek Water, Chemical Flush Fluids, and Stormwater
- **Emergency Response** – full-scale onsite in three weeks
- FF Foam Concentrate – Precursors, 5,000 $\mu\text{g/L}$ PFAS
- Multiple Contaminants (1,500 mg/L COD)
- Small Footprint
- Treatment Objective: 0.25 $\mu\text{g/L}$ sum of PFAS by TOP assay



Emergency Response

2 weeks
permitted



3 weeks
system
onsite



4 weeks
installed



Lab Scale

Ozofractionation – Case Study

Large volume high COD, high PFAS impacted wastewater

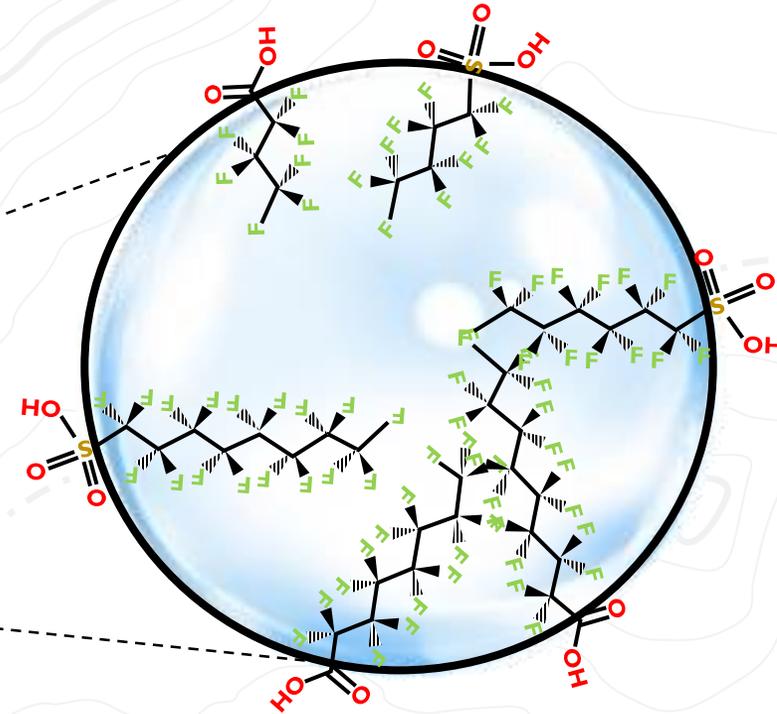
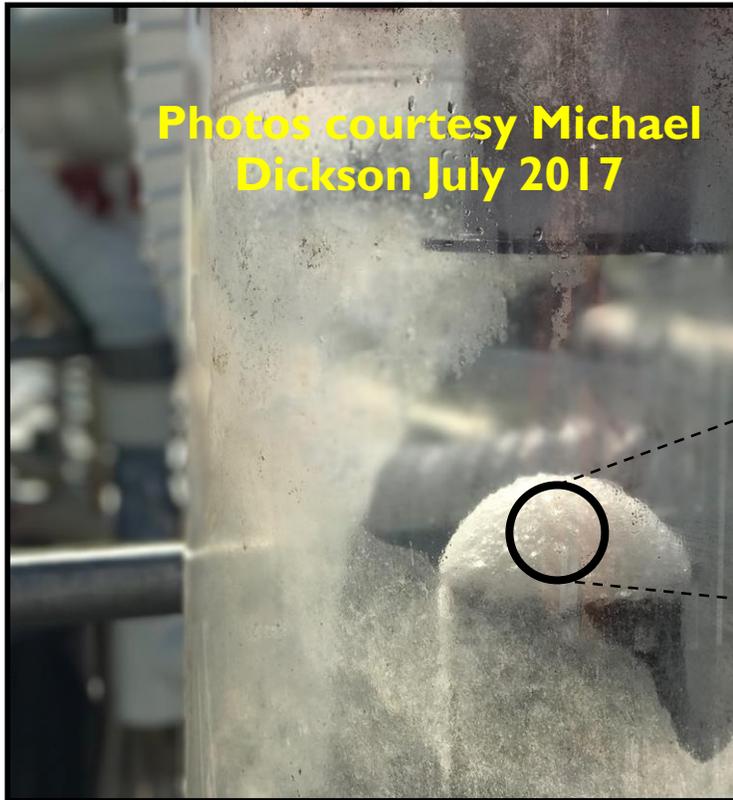
- ~3.6 million gallons of water
- Total [PFAS] ~ 3,950 µg/L; targeted discharge [PFAS] = <1 µg/L
- Laboratory analysis includes total oxidizable precursor (TOP) assay per country-specific regulations

Treatment train operation selected

- Ozofractionation with engineered polish
- Polish necessary for low discharge limit
- Foam concentrate to be thermally destroyed offsite



Ozofractionation - Concept



Ozofractionation – Case Study

Ozofractionation highly effective at removing PFOS, PFOA, and C6 PFAA precursors.

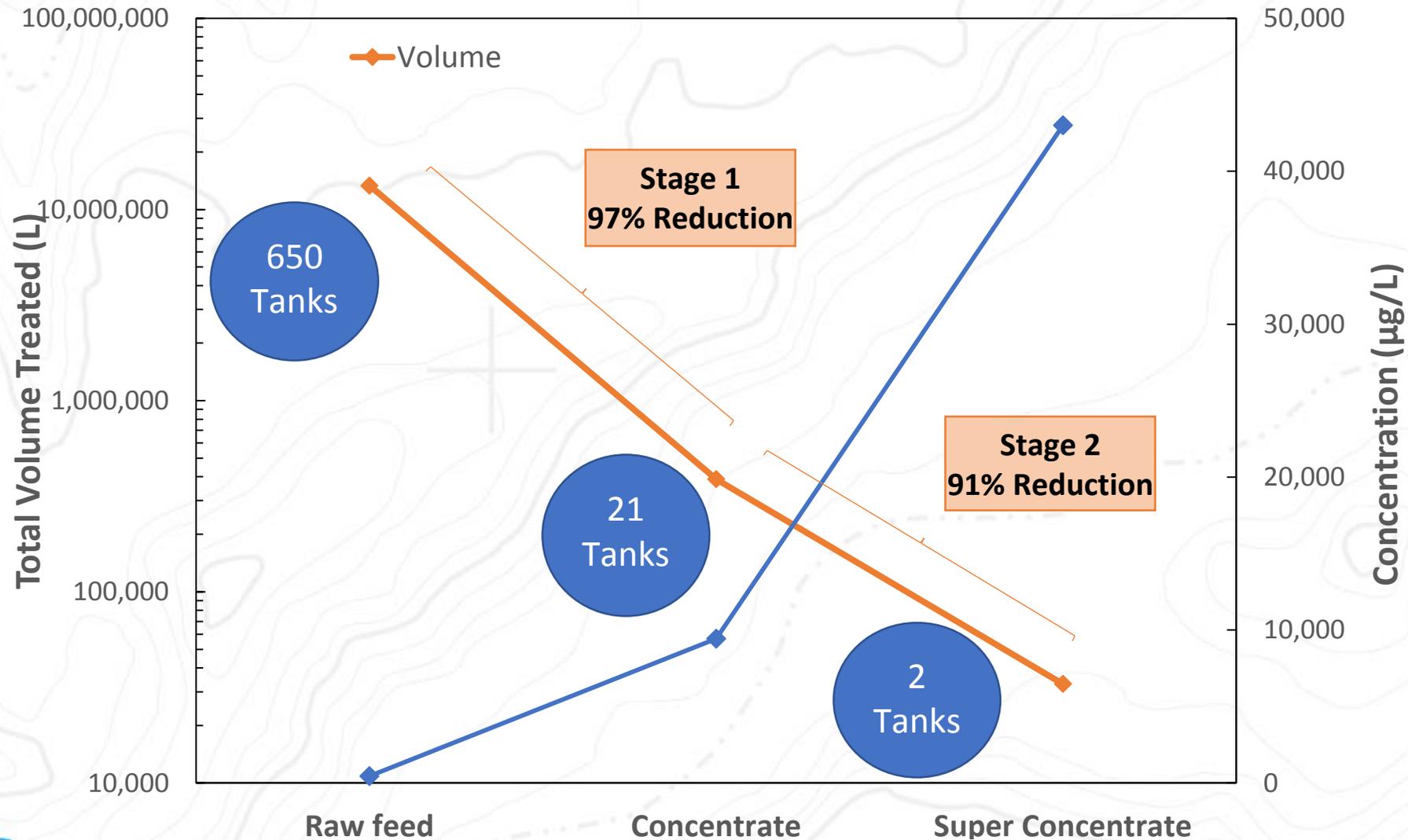
Ozofractionation converted some C6 precursors to PFHxA, PFPeA – net production of these compounds

Polishing adsorption stage was effective at removing PFHxA and, to a lesser extent, PFPeA; PFBA was not detectable in these samples

Identification	Influent (µg/L)	Ozofraction % Removal	Adsorbent % Removal	Treated Water (µg/L)	Total % Removal
PFOS	2.61	98.2%	81.3%	0.009	99.7%
PFOA	1.37	97.4%	94.4%	0.002	99.9%
6:2 FtS	87.4	95.6%	89.2%	0.416	99.5%
PFPeA	2.08	-66.3%	83.4%	0.575	72.4%
PFHxA	6.91	-66.4%	99.7%	0.034	99.5%
Sum PFAS	103	78.8%	95.1%	1.07	99.0%
Total PFAS, TOPA	3,950	99.6%	89.6%	1.76	99.96%

Ozofractionation and engineered polish achieve 99.96% PFAS removal, post TOP

Reconcentrate



Sonolysis

Applicability:

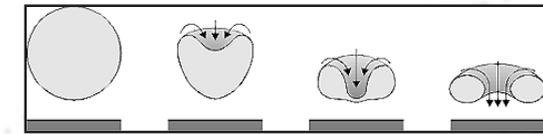
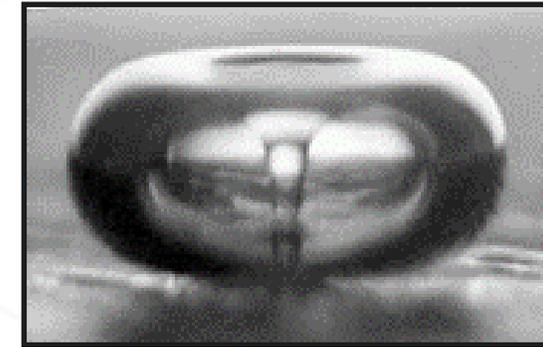
- Ultrasound applied to water results in successive rarefaction/compression of microbubbles ultimately yielding cavitation with extremely high temperatures on the surfaces of the bubbles resulting in pyrolysis of PFAS.

Benefits:

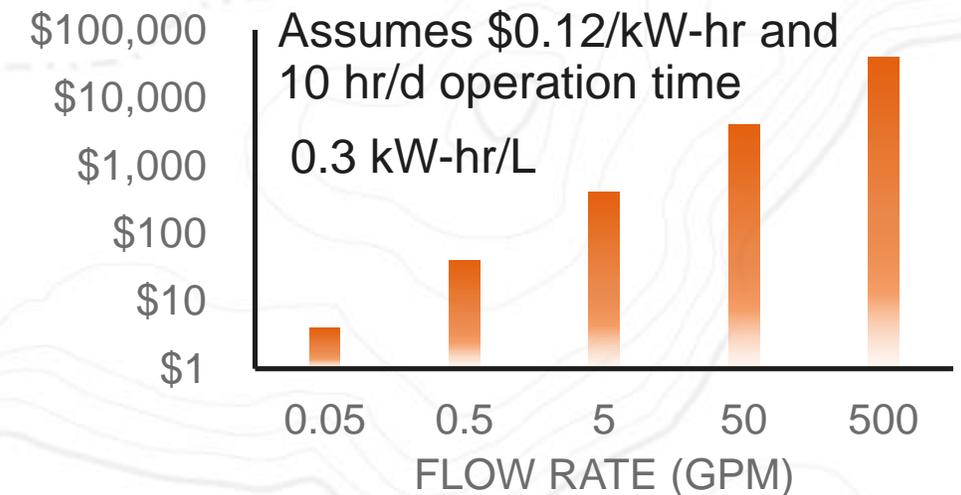
- Can reliably destroy concentrated PFAS waste streams with literature supported fluoride mass balance.
- Opportunities to use green energy sources as technology develops (i.e., solar power).

Limitations:

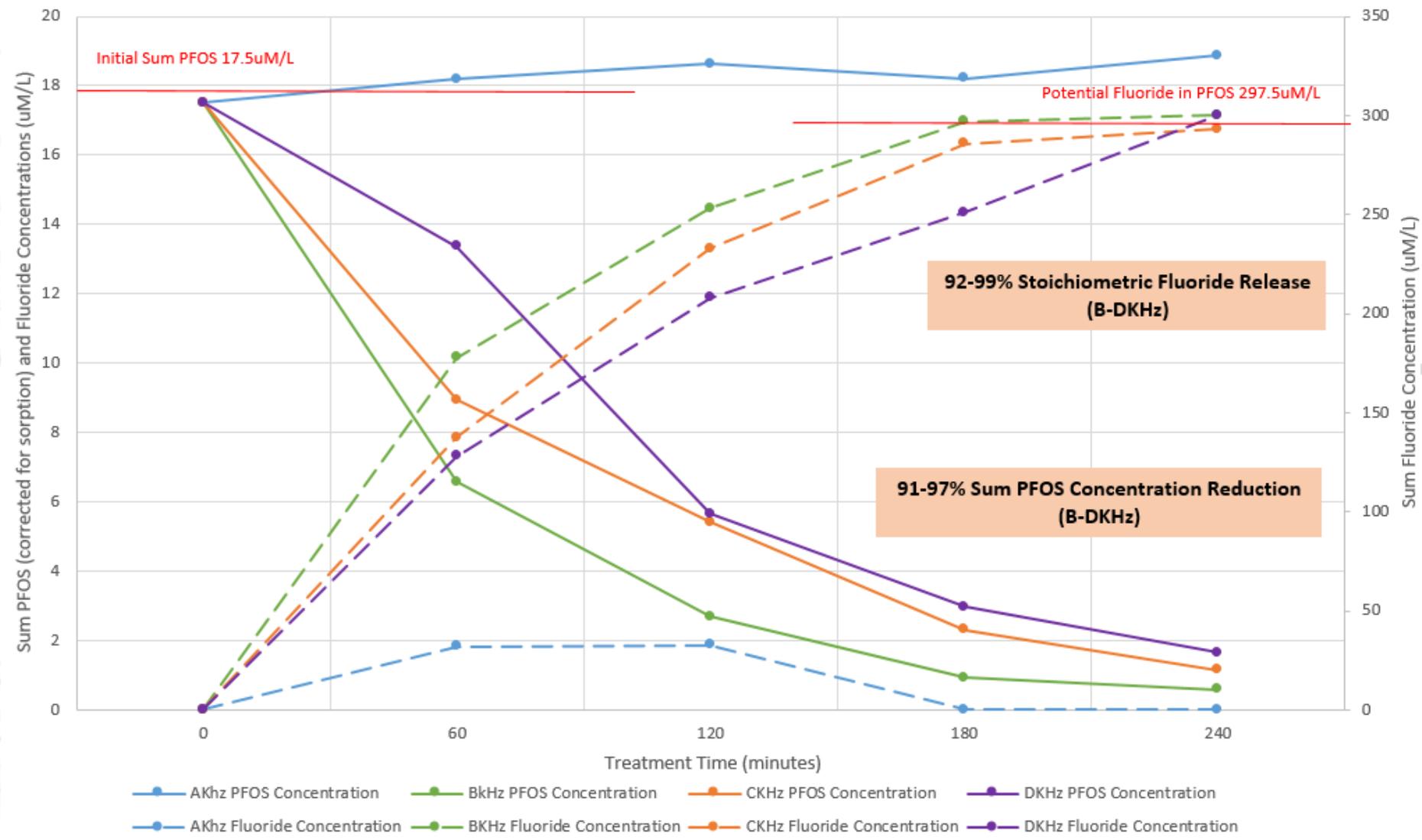
- PFOA rate > PFOS rate. PFOS will require longer residence times and/or more energy. Effective below 10,000 ppt?
- Requires specialized equipment and skilled implementation.
- High energy consumption and low flow rates.



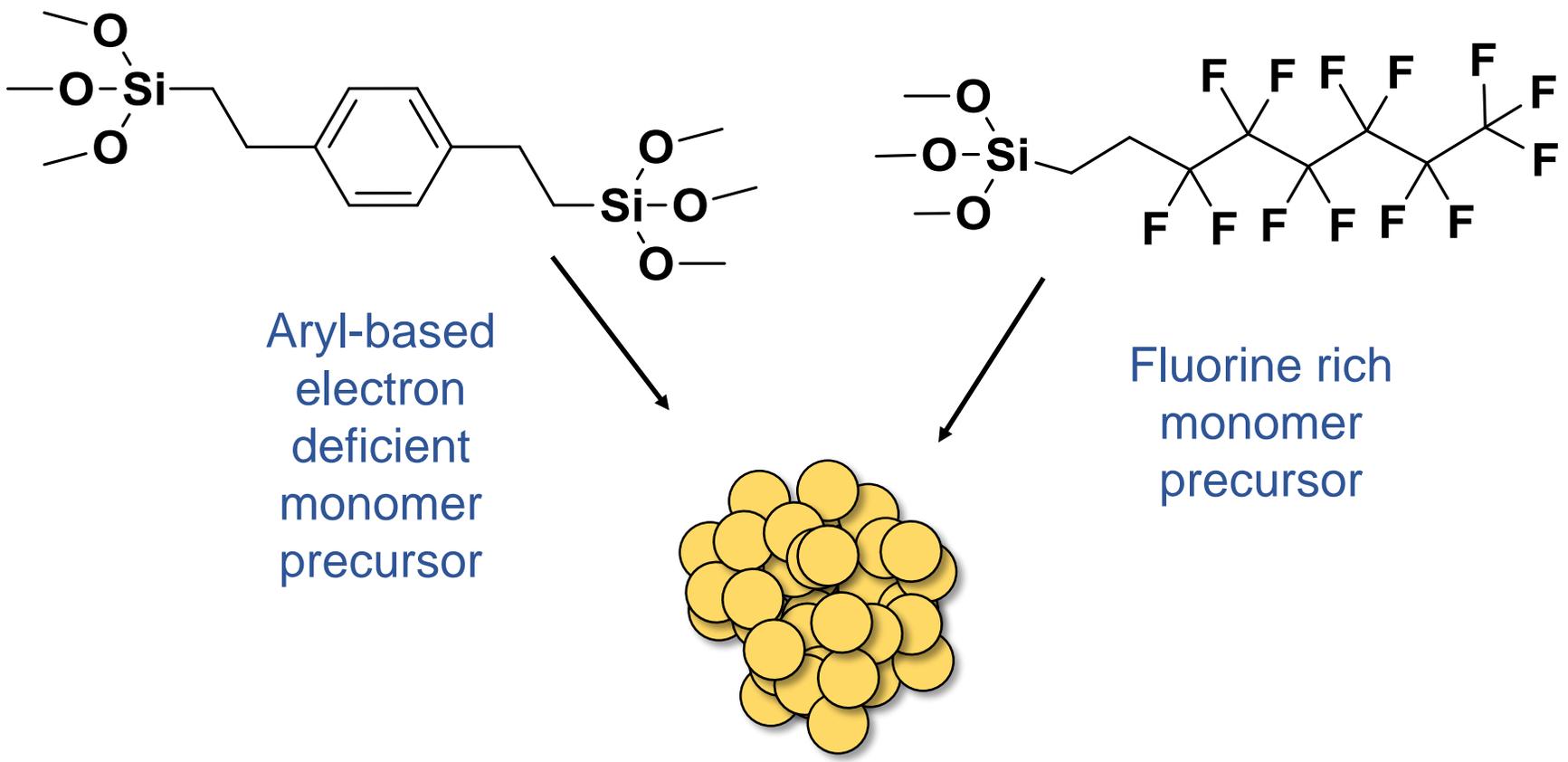
ENERGY COST (USD)



Sonolysis – Proof of Concept Testing



Engineered Regenerable Sorbents to Selectively Remove PFASs SERDP Research 2018

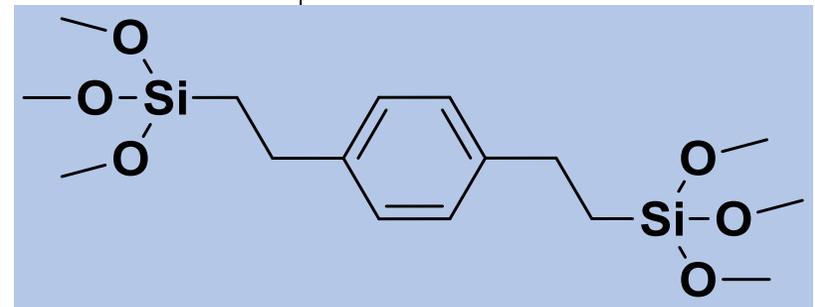
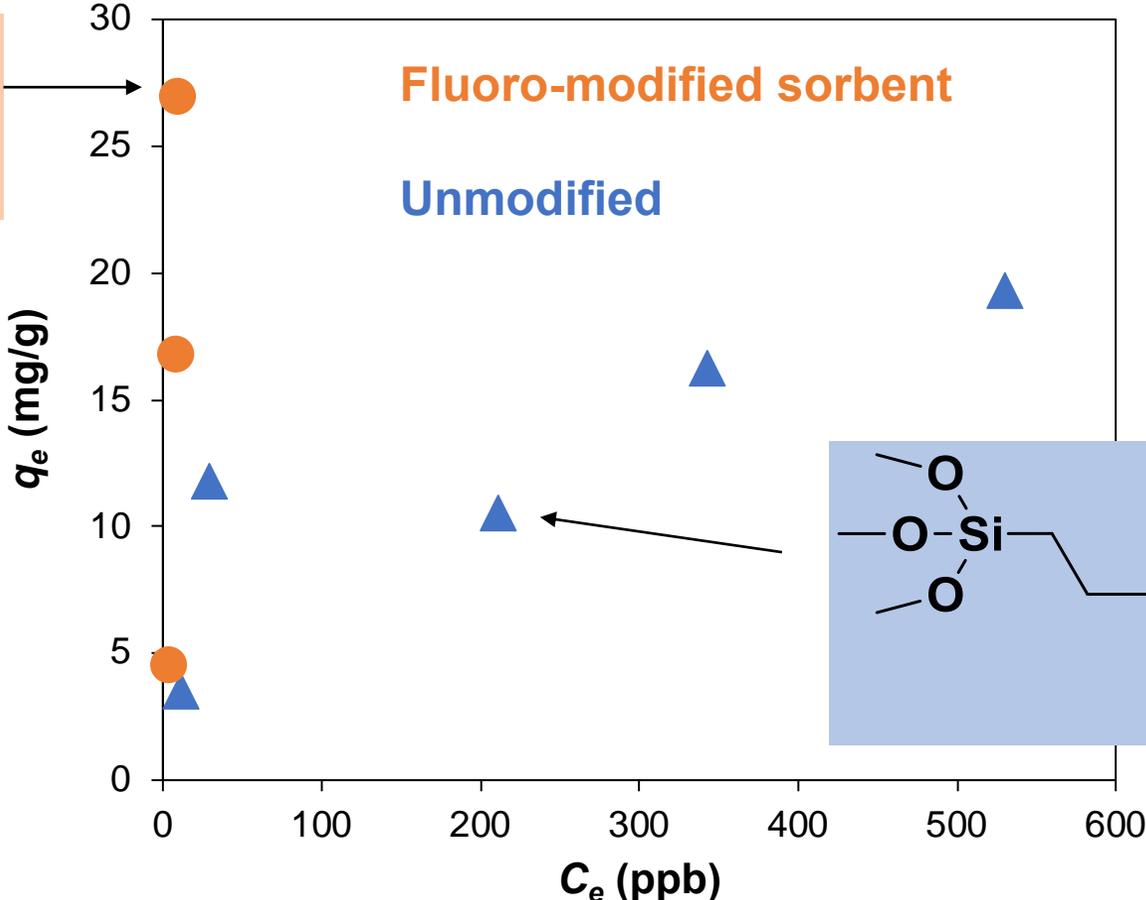
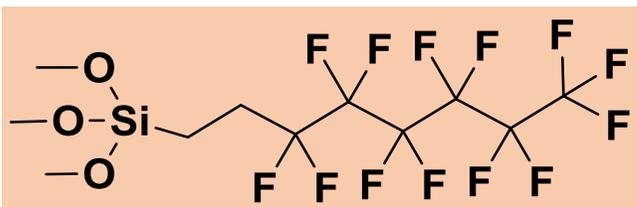


Silica-based porous solids with aryl and fluoro groups will lead to effective adsorption sites for PFASs by **interaction with the the fluoroalkyl chain.**

Engineered Regenerable Sorbents to Selectively Remove PFASs

SERDP Research 2018

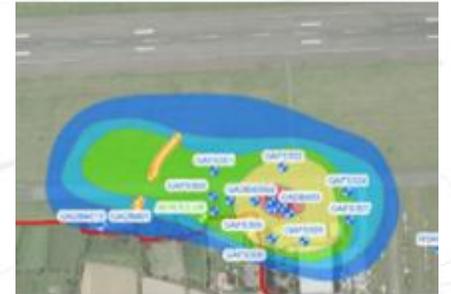
PFOS adsorption isotherm



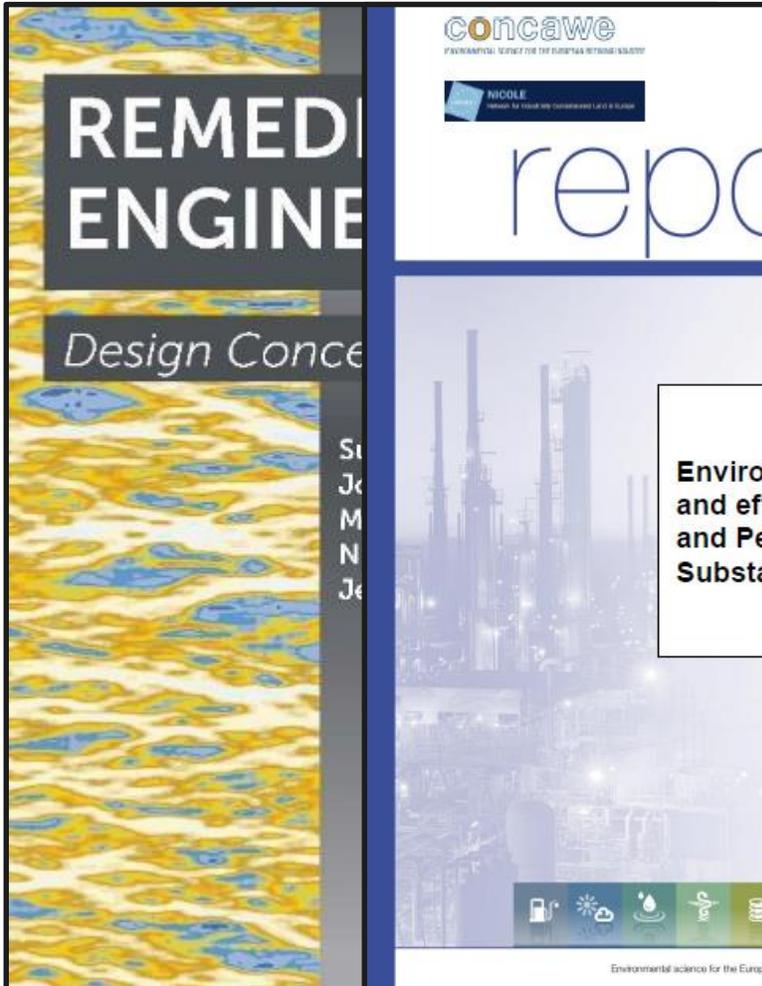
RESULTS: PFOS ADSORPTION ISOTHERM (90 BTEB / 10 TDF-TMS)

Summary

- Significant press attention & public concern on PFASs as a result of impacts to drinking water
- Significant decreases in drinking water standards globally
- Site assessment with TOP assay redefines the PFASs CSM
- How to apply the precautionary principle to be protective of human health & the environment, proportionate and sustainable?
- Risk based contaminated site management essential
- Emerging remedial technologies provide ingenious solutions for PFASs



Some Recent Publications



DOI: 10.1002/rem.21553

RESEARCH ARTICLE

WILEY

A review of emerging technologies for remediation of PFASs

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Abstract
The need for remediation of poly- and perfluoroalkyl substances (PFASs) is growing as a result of more regulatory attention to this new class of contaminants with diminishing water quality standards being promulgated, commonly in the parts per trillion range. PFASs comprise >3,000 individual compounds, but the focus of analyses and regulations has generally been PFASs termed perfluoroalkyl acids (PFAAs), which are all extremely persistent, can be highly mobile, and are increasingly being reported to bioaccumulate, with understanding of their toxicology evolving. However, there are thousands of polyfluorinated "PFAA precursors", which can transform in the environment and in higher organisms to create PFAAs as persistent daughter products.

Some PFASs can travel miles from their point of release, as they are mobile and persistent, potentially creating large plumes. The use of a conceptual site model (CSM) to define risks posed by specific PFASs to potential receptors is considered essential. Granular activated carbon (GAC) is commonly used as part of interim remedial measures to treat PFASs present in water. Many alternative treatment technologies are being adapted for PFASs or ingenious solutions developed. The diversity of PFASs commonly associated with use of multiple PFASs in commercial products is not commonly assessed. Remedial technologies, which are adsorptive or destructive, are considered for both soils and waters with challenges to their commercial application outlined. Biological approaches to treat PFASs report biotransformation which creates persistent PFAAs, no PFASs can biodegrade. Water treatment technologies applied *ex situ* could be used in a treatment train approach, for example, to concentrate PFASs and then destroy them on-site. Dynamic groundwater recirculation can greatly enhance contaminant mass removal via groundwater pumping. This review of technologies for remediation of PFASs describes that:

- GAC may be effective for removal of long-chain PFAAs, but does not perform well on short-chain PFAAs and its use for removal of precursors is reported to be less effective;
- Anion-exchange resins can remove a wider array of long- and short-chain PFAAs, but struggle to treat the shortest chain PFAAs and removal of most PFAA precursors has not been evaluated;

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